

CHEMISTRY



Atomic Structure

Atoms are the smallest particles that cannot be created or destroyed in a chemical reaction. They are made up of three sub-particles: electrons, protons, and neutrons.

Atomic Number

The atomic number of an element is the number of protons present in the nucleus of an atom of that element. It is denoted by the symbol Z .

Atomic Mass

The atomic mass of an element is the sum of the number of protons and neutrons present in the nucleus of an atom of that element. It is denoted by the symbol A .

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Element	Atomic Number	Atomic Mass
Hydrogen	1	1
Helium	2	4
Lithium	3	7
Boron	5	11
Carbon	6	12
Nitrogen	7	14
Oxygen	8	16
Fluorine	9	19
Neon	10	20

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