



Classification of Crystalline Compounds

Identifying Polymers, Solids, Minerals, and Liquids

Rank of Crystalline Compounds

When crystalline polymers, minerals, and solids are heated, they melt and become liquids. The melting point of a crystalline polymer is the temperature at which the polymer chains are able to move past each other. The melting point of a mineral is the temperature at which the mineral crystals are able to move past each other. The melting point of a solid is the temperature at which the solid is able to move past each other. The melting point of a liquid is the temperature at which the liquid is able to move past each other.

When you heat a crystalline polymer, mineral, or solid, you will see a change in its appearance. The solid will become a liquid and the liquid will become a gas. The change in appearance is due to the change in the arrangement of the particles. The particles in a solid are arranged in a regular pattern, while the particles in a liquid are arranged in a disordered pattern. The particles in a gas are arranged in a random pattern.

Questions: Rank the following crystalline compounds

1. Which compound is a crystalline polymer? Rank it according to its melting point.
2. Which compound is a crystalline mineral? Rank it according to its melting point.
3. Which compound is a crystalline solid? Rank it according to its melting point.
4. Which compound is a crystalline liquid? Rank it according to its melting point.
5. Which compound is a crystalline gas? Rank it according to its melting point.
6. Which compound is a crystalline solid? Rank it according to its melting point.
7. Which compound is a crystalline liquid? Rank it according to its melting point.
8. Which compound is a crystalline gas? Rank it according to its melting point.
9. Which compound is a crystalline solid? Rank it according to its melting point.
10. Which compound is a crystalline liquid? Rank it according to its melting point.

Identifying Organic Compounds Lab Answers

J Elliott



Identifying Organic Compounds Lab Answers:

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