

## Chapter 4 Atomic Structure

**Section 4.2 The Structure of an Atom****(pages 108–112)**

*This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.*

**Reading Strategy (page 108)**

**Monitoring Your Understanding** Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Know About Atoms	What I Would Like to Learn	What I Have Learned
Most students will know that atoms are the “building blocks” of matter, and some may know that atoms contain subatomic particles.	Based on the title of the section, students may say that they want to learn more about the structure of atoms.	

**Properties of Subatomic Particles (pages 108–109)**

- What are three subatomic particles?  
a. Protons      b. Electrons      c. Neutrons
- Circle the letter that identifies a subatomic particle with a positive charge.  
a. nucleus      **(b)** proton  
c. neutron      d. electron
- Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.  
\_\_\_\_\_  
\_\_\_\_\_

**Comparing Subatomic Particles (pages 109–110)**

- Circle the letters of properties that vary among subatomic particles.  
a. color      **(b)** mass  
**(c)** charge      **(d)** location in the atom
- Circle the letter of the expression that accurately compares the masses of neutrons and protons.  
**(a)** mass of 1 neutron = mass of 1 proton  
b. mass of 2000 neutrons = mass of 1 proton  
c. mass of 1 electron = mass of 1 proton  
d. mass of 1 neutron = mass of 1 electron

# Chapter 4 Atomic Structure Test A Answers

**Wolfgang Guggemos**



## Chapter 4 Atomic Structure Test A Answers:

*New York State Regents Exam* Kaplan, Kaplan Educational Center Staff, 2000-02 Kaplan's guides to the New York State Regents Exams come complete with a comprehensive review of all the tested material plus Kaplan's exclusive test taking strategies. This powerful combination makes the *New York State Regents Exam Chemistry Second Edition* a highly effective way for you to score higher on this very challenging test. Are you ready for the New York State Regents Chemistry exam? You will be with Kaplan's proven plan for success. STEP 1 Take a Diagnostic Test. The results of this test will outline your strengths and weaknesses. You will find out the exact areas on which you need to focus your preparation. STEP 2 Review the Tested Material. Kaplan takes you through each section step by step providing you with effective tips and strategies to successfully answer every type of question. STEP 3 Practice with Real Regents Exams. Practice makes perfect. And with this book you'll practice with the real thing: actual Regents exams including detailed explanations for every answer and an analysis of your performance. STEP 4 Succeed on the Test. Follow Kaplan's plan for success on the Regents and you will score higher. In fact we guarantee it. See details inside. **McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition** Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam. Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects. McGraw Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need to perform at your best. This book includes 4 full length sample tests updated for the latest test format. 40 top tips to remember on test day. Glossary of tested chemistry terms and formulas. Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School. Diagnostic test to pinpoint strengths and weaknesses. Step by step review of all topics covered on the exam. In depth coverage of the lab experiment questions that are a major test feature. Charts, tables and illustrations to simplify and reinforce learning. Practice tests just like the real SAT Subject Test in Chemistry. Test taking tips and strategies.

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