

Name _____ Date _____ Class _____

CHAPTER

15

STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

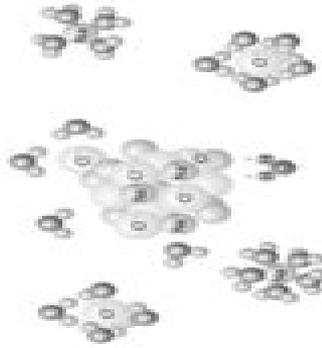
Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solute
miscible	insoluble	solvent	solvent

Air is a(n) **(1)** solution of oxygen gas dissolved in nitrogen gas. The oxygen is an **(2)** solute, and nitrogen is the **(3)** solvent. Because oxygen gas dissolves in a solvent, oxygen gas is an **(4)** soluble substance. A substance that does not dissolve is **(5)** insoluble. **(6)** Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** miscible. However, if one liquid is insoluble in another, the liquids are **(8)** immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



9. Hydration is hydration in which the solvent is water. What is hydration?
Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER

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Section 15.1 continued

10. An sodium chloride dissolves in water. What happens to the sodium and chloride ions? **The sodium and chloride ions are separated and surrounded by the water molecules.**

water molecules.

11. Explain the orientation of the water molecules around the sodium ions and chloride ions. **Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.**

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? **How do you know?**

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write *true* or *false*.

- true** 14. The overall energy change that occurs when a solution forms is called the heat of solution.
- false** 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.
- false** 16. Solvation continues as long as the solvation rate is less than the crystallization rate.
- true** 17. In a saturated solution, solvation and crystallization are in equilibrium.
- true** 18. Additional water can be dissolved in an unsaturated solution.
- true** 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Ann Johnson, Richard A. Kalish



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chapters offer suggested readings based on student participation in the skills lab and discussions with instructor Audio visual assignments with related Critical Thinking Questions tie the visual video demonstrations of specific examination procedures to practical applications in the skills lab Coverage of the electronic health record charting and narrative recording gives students examples of how to document assessment findings Anatomy labelling exercises visually reinforce the identification of key anatomy and physiology Study guide activities reinforce key assessment information through short answer fill in the blank and labelling exercises Reading assignments correspond to the text chapters to foster integration of the text and laboratory manual Documentation Sheets allow students and faculty to assess knowledge with forms used in the skills lab or clinical setting Current RN and PN competencies are reflected throughout to help students better prepare for licensure examinations Review questions short answer matching multiple choice provide learning activities in a variety of approaches Narrative Summary Forms reflect the charting format used for narrative accounts of the history and physical examination findings Glossary promotes learning and understanding of essential terminology Answer key on the companion Evolve website includes answers to all questions NEW Preface includes an overview of the NCSBN Clinical Judgement Measurement Model PN competencies and the REx PN NEW Clinical judgement exercises equip students for success on the Next Generation NCLEX NGN including questions with an increased focus on clinical judgement Updated content throughout corresponds to the 4th Canadian edition of the textbook and incorporates the latest research and evidence informed practice

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answer matching multiple choice and answer key Chapter topics include 1 an overview of the human body 2 the problem of food choices 3 nutrient needs and nutrition surveys 4 energy balance overweight and underweight 5 food composition and safety and 6 measures for improving one's diet The following nutrients are examined in detail carbohydrates sugar starch fiber lipids proteins vitamins minerals and water Nutrient requirements throughout the life cycle i e pregnancy infancy childhood adolescence adulthood are also reviewed Supplementary topics include the role of doctors in nutrition the appropriateness of dietary guidelines nutrition and acne and nutrition and cancer Appendices list reliable and unreliable sources of nutrition information and present a series of self study exercises and accompanying forms for evaluating individual dietary and exercise habits aje

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