

**UNIT 10: PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

**Section 10.1** *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____
_____

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                                  |
|---|----------------------------------|
| _____ 18. A column in the periodic table                  | _____ a. metals                  |
| _____ 19. A row in the periodic table                     | _____ b. group                   |
| _____ 20. Group 1 elements                                | _____ c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements                                | _____ e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- \_\_\_\_\_ 29. Alkaloids have properties of both metals and more reactive metals.

# Chapter 6 Study Guide For Content Mastery Answers

**Garin Hess**



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