

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
- how many atoms of Fe are in one formula unit
 - how many sulfate ions can be attached to the iron atom
 - the charge on each Fe ion
 - the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to _____
- change the number of moles represented by the formula
 - change the charges on the other ions in the compound
 - change the formula so that it no longer represents the compound it previously represented
 - have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
- | | |
|--|---|
| <u>4 elements</u> | a. How many elements make up this compound? |
| <u>6 oxygen atoms</u> | b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u>21 atoms</u> | c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u>4.2×10^{24} atoms</u> | d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$? |
4. How many atoms are present in each of these formula units?
- | | |
|-----------------|-------------------------|
| <u>11 atoms</u> | a. $Ca(HCO_3)_2$ |
| <u>45 atoms</u> | b. $C_{12}H_{22}O_{11}$ |
| <u>10 atoms</u> | c. $Fe(ClO_2)_3$ |
| <u>9 atoms</u> | d. $Fe(ClO_3)_3$ |
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
- carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
- H_2SO_3 c. What is the formula for sulfurous acid?
- phosphoric acid d. What is the name for the acid H_3PO_4 ?

Chapter 7 Chemical Formulas And Compounds Test

Oregon. Board of Horticulture



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A Method for the Identification of Pure Organic Compounds by a Systematic Analytical Procedure Based on Physical Properties and Chemical Reactions ... Samuel Parsons Mulliken,1904 **Chemical Interactions** ,2005 **Report** Oregon. Board of Horticulture,1917 *Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed* Peter P. Berlow,1978 **2024-25 CBSE/NIOS/ISC/UP Board 12th Class Chemistry Chapter-wise Unsolved Papers** YCT Expert Team , 2024 25 CBSE NIOS ISC UP Board 12th Class Chemistry Chapter wise Unsolved Papers 464 895 E This book contains the previous year paper from 2010 to 2024 *A Method for the identification of pure organic compounds by a systematic analytical procedure based on physical properties and chemical reactions ... v. 1, 1911* Samuel Parsons Mulliken,1904 *Heath Science Connections 9* Jack L. Candido,Edward S. James,Ronald E. Phillips,1987 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *College Study Skills* Dianna L. Van Blerkom,2000 This comprehensive text workbook focuses on getting students actively involved in the learning process and in learning how to learn Rather than presenting one prescriptive approach students take a learning style inventory and are then exposed to strategies that complement their personal learning styles **The Systematic Identification of Organic Compounds** Ralph L. Shriner,1998 Dedicated to qualitative organic chemistry this book explains how to identify organic compounds through step by step instructions Topics include elemental analysis solubility infrared nuclear magnetic resonance and mass spectra classification tests and preparation of a derivative Most directions for experiments are described in micro or mini scales

Discusses chromatography distillations and the separation of mixtures Questions and problems emphasize the skills required in identifying unknown samples **Biennial Report** Oregon. Board of Horticulture, 1917 **Philadelphia Monthly Medical Journal** , 1899

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