

AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which pair of atoms should form the most polar bond?
a. F and B
b. C and O
c. F and O
d. N and F
e. B and N
- _____ 2. Which pair of ions should form the ionic lattice with the highest energy?
a. Na^{1+} and Br^{1-}
b. Li^{1+} and F^{1-}
c. Cs^{1+} and F^{1-}
d. Li^{1+} and O^{2-}
e. K^{1+} and F^{1-}
- _____ 3. The electrostatic force of attraction is greatest in which compound?
a. BaO
b. MgO
c. CaS
d. MgS
e. CaO
- _____ 4. Which molecule has the weakest bond?
a. CO
b. O_2
c. NO
d. N_2
e. Cl_2
- _____ 5. For which species are octet resonance structures necessary and sufficient to describe the bonding satisfactorily?
a. BCl_3
b. SO_2
c. CO_2
d. BeF_2
e. SO_4^{2-}
- _____ 6. How are the bonding pairs arranged in the best Lewis structure for ozone?
a. $\text{O}-\text{O}-\text{O}$
b. $\text{O}=\text{O}-\text{O}$
c. $\text{O}=\text{O}=\text{O}$
d. $\text{O}=\text{O}=\text{O}$
e. $\text{O}=\text{O}=\text{O}$
- _____ 7. Which species has the shortest bond length?
a. CN^{1-}
b. O_2
c. SO_2
d. SO_3
e. CO_2
- _____ 8. Which species requires the least amount of energy to remove an electron from the outermost energy level?
a. Na^{1+}
b. He
c. F^{1-}
d. O^{2-}
e. Mg^{2+}
- _____ 9. Which species has a valid non-octet Lewis structure?
a. GeCl_4
b. SiF_4
c. CO_3^{2-}
d. SeCl_4
e. CO_3^{2-}

Chapter 8 Chemistry Answers

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Study Guide, Answer Book to Accompany Organic Chemistry Alan S. Wingrove, Robert L. Caret, 1981

British Museum Catalogue of printed Books, 1897

Journal of the Franklin Institute Franklin Institute (Philadelphia, Pa.), 1956 Vols 1-69 include more or less complete patent reports of the U.S. Patent Office for years 1825-1859 cf Index to v 1-120 of the Journal p 415

Answer Book to Accompany Boikess/Edelson, Chemical Principles, Third Edition Robert S. Boikess, 1985

McGraw-Hill's New MCAT George J. Hademenos, 2007 This title is more focused and student friendly than competing MCAT guides It has more special features including a companion CD with two full length timed practice MCATs and free website support There is more bang for the buck this is a complete all in one preparation package for the 60,000 students who take the MCAT every year

General Science Quarterly, 1924

Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines

Study Guide with Solutions for Organic Chemistry, Third Edition Ralph J. Fessenden, Joan S. Fessenden, 1986 Provides a set of additional drill problems chapter by chapter discussions and supplemental instructional material to help students master organic chemistry problem solving techniques

MCAT Comprehensive Review, 2004

Science Explorer Physical Science Michael J. Padilla, Ioannis Miaculis, Martha

Cyr, **Preparation for the Upper Division** Kathleen A. Hart, 1994

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