

CHAPTER 4 The Structure of Matter

SECTION

2

Ionic and Covalent Bonding

KEY IDEAS

As you read this section, keep these questions in mind:

- Why do atoms form bonds?
- How do ionic bonds and covalent bonds differ?
- What gives metals their distinctive properties?
- What makes a polyatomic ion different from other ions?

Why Do Chemical Bonds Form?

Atoms form bonds when their valence electrons interact. Recall that atoms with filled outermost energy levels are more stable than atoms with partially filled energy levels. In general, atoms join to form bonds so that each atom has a stable electron configuration. In other words, each atom binds in order to fill its outermost energy levels. \square

There are two basic kinds of chemical bonding: ionic bonding and covalent bonding. The type of bonding in a compound determines many of the properties of the compound.

A Comparison of Ionic and Covalent Compounds		
	Ionic compounds	Covalent compounds
Structure	network of bonded ions	molecules
Valence electrons	transferred	shared
Electrical conductivity	good (when melted or dissolved)	poor
State at room temperature	solid	solid, liquid, or gas
Melting and boiling points	generally high	generally low

What Are the Properties of Ionic Bonds?

Ionic bonds form between oppositely charged ions. In general, atoms of metals, such as sodium and calcium, form positively charged ions. Atoms of nonmetals, such as chlorine and oxygen, form negatively charged ions. The attraction between ions with opposite charges holds ionic compounds together. \square

READING TOOLBOX

Compare As you read, make a chart that describes and compares the different types of bonds.

READING CHECK

1. **Explain** Why do atoms join to form bonds?

LOOKING CLOSER

2. **Compare** How do the structures of ionic and covalent compounds differ?

READING CHECK

3. **Identify** What holds ionic compounds together?

Chapter Section 2 Ionic And Covalent Bonding

KJ Lindholm-Leary



Chapter Section 2 Ionic And Covalent Bonding:

The Top Books of the Year Chapter Section 2 Ionic And Covalent Bonding The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the engaging narratives that have enthralled audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and suspenseful novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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