

Chapter Test B

A. Matching

Match each item in Column A with the correct description in Column B. Write the letter of the correct description.

Column A

1. An unbalanced chemical equation for a reaction that obeys the law of conservation of matter.
2. The number of protons in an atom of an element. It is equal to the number of electrons.
3. The number of atoms of an element in a chemical formula.
4. The ratio of the actual yield to the theoretical yield, expressed as a percent.
5. The substance that is present in excess in a chemical reaction.
6. An equation whose coefficients are the smallest whole numbers possible.

Column B

- a. percent yield
- limiting reagent
- theoretical yield
- stoichiometry
- excess reagent
- balanced equation

B. Multiple Choice

Choose the best answer for each question.

1. Which of the following is not a chemical reaction?
 - a. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - $\text{NaCl} \rightarrow \text{Na} + \text{Cl}_2$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$
2. Which of the following is not a chemical reaction?
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
3. Which of the following is not a chemical reaction?
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$

Chemistry Chapter 12 Stoichiometry Test

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