

Chemistry
Chapter 9 Practice Test: Stoichiometry

Name _____

Free Response

Use the following equation to answer problems 1 - 4 at STP. (Show All Work)



1. How many **moles** of aluminum chloride (AlCl_3) would be produced from 0.12 moles of Hydrochloric acid (HCl)?
2. If you wanted to completely react 5.0 moles of Aluminum, how many **moles** of hydrochloric acid (HCl) would be needed?
3. How many **grams** of hydrochloric acid (HCl) would be needed to completely react 35.8 g of Aluminum foil?
4. How many **grams** of aluminum would be needed to produce 67.2 dm³ of hydrogen gas?

Use the following equation to answer problems 5 and 6 at STP.



5. How many **molecules** of carbon dioxide (CO_2) would be produced by completely reacting 8.00 grams of oxygen gas (O_2) with excess Carbon Monoxide (CO)?
6. What **volume** in L of oxygen gas (O_2) would react with 6.0 L of carbon monoxide gas (CO) ?

Chemistry Chapter 9 Stoichiometry Test

Robert Bruce Thompson



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