

**Equilibrium**

- a. The partial \_\_\_\_\_ of a diatomic gas \_\_\_\_\_ of gas varies inversely \_\_\_\_\_ with the \_\_\_\_\_ of that gas \_\_\_\_\_.
- b. \_\_\_\_\_.
- c. \_\_\_\_\_.

**Gas Theory and Equilibrium**

- a. \_\_\_\_\_ of a gas is directly related \_\_\_\_\_ of the gas \_\_\_\_\_ the \_\_\_\_\_ of the \_\_\_\_\_.
- b. If the gas is compressed by \_\_\_\_\_ the volume of the \_\_\_\_\_ is \_\_\_\_\_ in every \_\_\_\_\_, \_\_\_\_\_ pressure is \_\_\_\_\_.
- c. \_\_\_\_\_ in every \_\_\_\_\_, \_\_\_\_\_ pressure is \_\_\_\_\_.
- d. \_\_\_\_\_ of the \_\_\_\_\_.
- e. \_\_\_\_\_ of the \_\_\_\_\_.

**Q10** A certain diatomic gas \_\_\_\_\_ of \_\_\_\_\_ at \_\_\_\_\_ pressure \_\_\_\_\_ after the volume is \_\_\_\_\_, its \_\_\_\_\_ is \_\_\_\_\_ to be \_\_\_\_\_ the pressure is only \_\_\_\_\_ than is a \_\_\_\_\_ of the \_\_\_\_\_ in this situation?

\_\_\_\_\_

\_\_\_\_\_

**Q11** Under conditions of \_\_\_\_\_, \_\_\_\_\_ is a \_\_\_\_\_ gas \_\_\_\_\_ of \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_ pressure is \_\_\_\_\_ of the \_\_\_\_\_ in \_\_\_\_\_, \_\_\_\_\_ is a \_\_\_\_\_ assuming the \_\_\_\_\_ remains \_\_\_\_\_.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

# Chemistry Episode 902 Note Taking Guide Answers

**Lauren Gardner**



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