

**Equilibrium**

- **Equilibrium** is a state where the net force on an object is zero. It can be static or dynamic.
- **Static Equilibrium**: The object is at rest.
  - $\sum F = 0$
  - $\sum \tau = 0$
- **Dynamic Equilibrium**: The object is moving with constant velocity.
  - $\sum F = 0$
  - $\sum \tau = 0$
- **Center of Mass**: The point where the mass of the object is concentrated.
  - $\sum \tau = 0$  if the center of mass is directly above the pivot.
- **Stable Equilibrium**: The object returns to its original position after a small displacement.
- **Unstable Equilibrium**: The object moves away from its original position after a small displacement.
- **Neutral Equilibrium**: The object remains in its original position after a small displacement.



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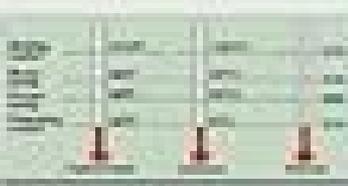
**Nuclear Chemistry**

- **Radioactivity** is the process by which unstable atomic nuclei lose energy by emitting radiation.
- **Types of Radiation**:
  - **Alpha** ( $\alpha$ ): Helium nucleus,  $^4_2\text{He}$
  - **Beta** ( $\beta$ ): Electron,  $^0_{-1}\text{e}$
  - **Gamma** ( $\gamma$ ): High-energy photon,  $^0_0\gamma$
- **Half-life** ( $t_{1/2}$ ): The time it takes for half of the radioactive substance to decay.
- **Radioactive Decay**:
  - $N = N_0 e^{-\lambda t}$
  - $t_{1/2} = \frac{\ln 2}{\lambda}$
- **Applications**:
  - **Medicine**: Cancer treatment, diagnostic imaging.
  - **Industry**: Radiography, smoke detectors.
  - **Energy**: Nuclear power plants.



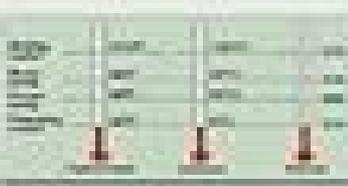

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- **SI Units**: International System of Units.
- **Base Units**:
  - Length: meter (m)
  - Mass: kilogram (kg)
  - Time: second (s)
  - Temperature: kelvin (K)
  - Amount of substance: mole (mol)
  - Electric current: ampere (A)
  - Luminous intensity: candela (cd)
- **Derived Units**:
  - Area: square meter ( $\text{m}^2$ )
  - Volume: cubic meter ( $\text{m}^3$ )
  - Density:  $\text{kg}/\text{m}^3$
  - Speed:  $\text{m}/\text{s}$
  - Acceleration:  $\text{m}/\text{s}^2$
  - Force: newton (N)
  - Energy: joule (J)
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