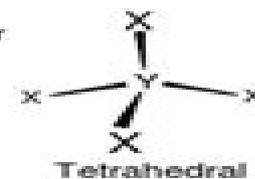
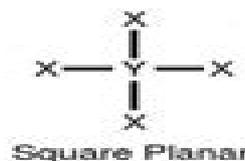
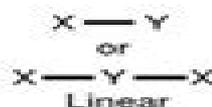


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Determine if the molecules listed below are polar or nonpolar. To be polar, a molecule must:

- *have polar-covalent bonds AND*
- *not be symmetrical.*

Note: see tables on pages 177 and 238

Oxygen gas, O₂

Electronegativities: O O Difference Polar or nonpolar bond?

Electron dot diagram Structural formula Shape Polar or Nonpolar Molecule?

Nitrogen gas, N₂

Electronegativities: N N Difference Polar or nonpolar bond?

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H₂CS

Electronegativities: N N Difference Polar or nonpolar bond?

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Ammonium ion, NH₄⁺

Electronegativities: N N Difference Polar or nonpolar bond?

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