

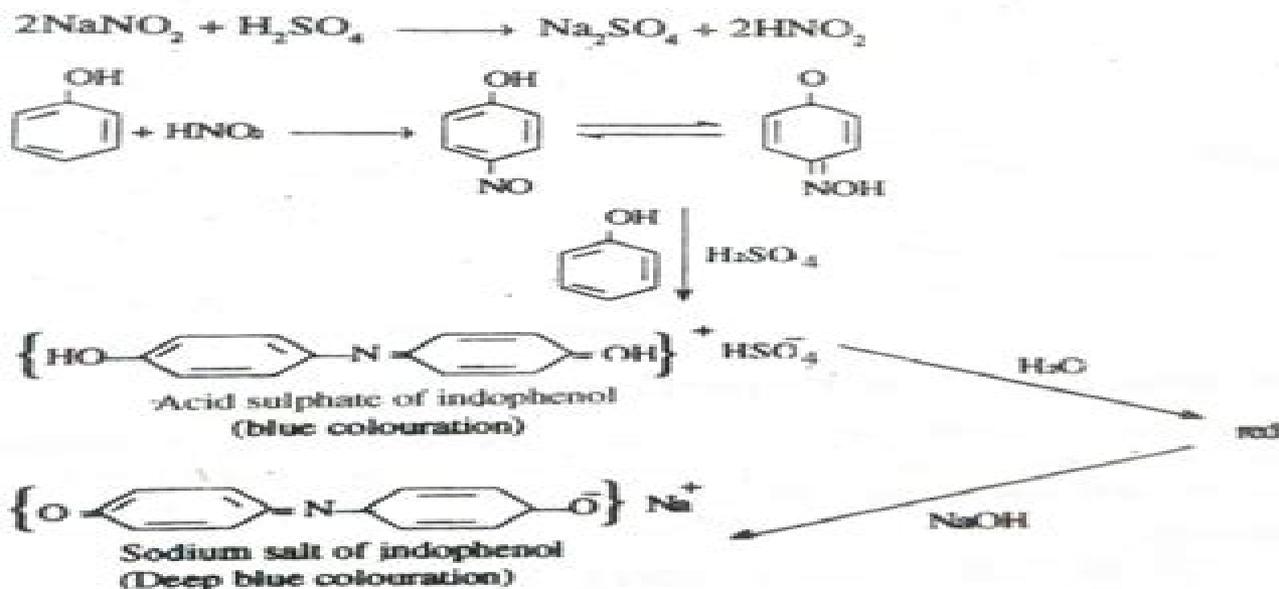
5.2 PHENOLIC GROUP (ArOH)

Phenols are colourless but attain colour due to oxidation. Moreover, they are acidic and dissolve in 5% NaOH. Phenols are generally highly reactive in electrophilic aromatic substitution. On bromination (bromine water) of phenol, 2, 4, 6 - tribromophenol is obtained.

- (i) Litmus paper test (ii) NaHCO_3 test (iii) FeCl_3 test
 (iv) Bromine water test (v) Azodye test
 (vi) Liebermann nitroso reaction

Dissolve 0.2 g of the substance in 2 mL conc. H_2SO_4 and add a few crystals of NaNO_2 — a blue colouration develops immediately which becomes red on dilution with water. Add NaOH in excess. The colour changes to green, blue or violet. Among dihydroxyphenols only resorcinol gives a satisfactory positive test.

Chemical reaction



Only those phenols possessing a free para position respond to this test.

Chemistry Test Functional Groups Chapter 23

**Thirumalaisamy P.
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