

1E1004

Roll No. _____

Total No of Pages: **3****1E1004**

**B. Tech I Sem. (Back) Exam. Jan. 2016
203 Old Engineering Chemistry
Common to all Branches of Engineering**

Time: 3 Hours**Maximum Marks: 80
Min. Passing Marks: 24***Instructions to Candidates:*

Attempt overall five questions, selecting one question from each unit. All questions carry equal marks.

*Use of following supporting material is permitted during examination.
(Mentioned in form No. 205)*

1. NIL2. NIL**UNIT-I**

- Q.1 (a) What is EDTA? How will you determine the hardness of water by EDTA? [3+3=6]
- (b) What is Break point chlorination? [4]
- (c) Water containing CaCO_3 120 mg/lit, MgCO_3 84 mg/lit, $\text{Mg}(\text{NO}_3)_2$ 74 mg/lit, CaCl_2 111 mg/lit and KCl 100 mg/lit is to be treated with lime soda. Calculate the amount of chemicals required for the treatment of 5000 lit of water. [6]

OR

- Q.1 (a) Describe lime soda process for water softening. Give the chemical reactions involved during the softening. [3+3=6]
- (b) If silica is present in water, what harmful effects it can cause? [4]
- (c) Calculate the amount of lime required for softening of 5000 lit of hard water containing 72 ppm of Mg SO_4 . [6]

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Jianjun Gao



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Table of Contents Engineering Chemistry Question Paper May 2016 First

1. Understanding the eBook Engineering Chemistry Question Paper May 2016 First
 - The Rise of Digital Reading Engineering Chemistry Question Paper May 2016 First
 - Advantages of eBooks Over Traditional Books
2. Identifying Engineering Chemistry Question Paper May 2016 First
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Engineering Chemistry Question Paper May 2016 First
 - User-Friendly Interface
4. Exploring eBook Recommendations from Engineering Chemistry Question Paper May 2016 First

- Personalized Recommendations
 - Engineering Chemistry Question Paper May 2016 First User Reviews and Ratings
 - Engineering Chemistry Question Paper May 2016 First and Bestseller Lists
5. Accessing Engineering Chemistry Question Paper May 2016 First Free and Paid eBooks
- Engineering Chemistry Question Paper May 2016 First Public Domain eBooks
 - Engineering Chemistry Question Paper May 2016 First eBook Subscription Services
 - Engineering Chemistry Question Paper May 2016 First Budget-Friendly Options
6. Navigating Engineering Chemistry Question Paper May 2016 First eBook Formats
- ePub, PDF, MOBI, and More
 - Engineering Chemistry Question Paper May 2016 First Compatibility with Devices
 - Engineering Chemistry Question Paper May 2016 First Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Engineering Chemistry Question Paper May 2016 First
 - Highlighting and Note-Taking Engineering Chemistry Question Paper May 2016 First
 - Interactive Elements Engineering Chemistry Question Paper May 2016 First
8. Staying Engaged with Engineering Chemistry Question Paper May 2016 First
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Engineering Chemistry Question Paper May 2016 First
9. Balancing eBooks and Physical Books Engineering Chemistry Question Paper May 2016 First
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Engineering Chemistry Question Paper May 2016 First
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Engineering Chemistry Question Paper May 2016 First
- Setting Reading Goals Engineering Chemistry Question Paper May 2016 First
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Engineering Chemistry Question Paper May 2016 First

- Fact-Checking eBook Content of Engineering Chemistry Question Paper May 2016 First
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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