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## Collision Theory Worksheet

- 1) Explain why all reactions have an activation energy, using your knowledge of collision theory.
- 2) Describe how the activation energy of a reaction affects the overall rate of the chemical reaction.
- 3) A rule of thumb used by organic chemists is that the rate of a chemical reaction can be doubled by increasing the reaction temperature by ten degrees Celsius. Explain this drastic increase in reaction rate using your knowledge of collision theory.
- 4) *[Faint, illegible text]*
- 5) *[Faint, illegible text]*

# Collision Word Problem And Answers Soundmetals

**Scott C. Dulebohn**



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