

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	Very	Very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (III) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acids Bases Section 1 Answers

Karin Nielsen-Saines



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Table of Contents Acids Bases Section 1 Answers

1. Understanding the eBook Acids Bases Section 1 Answers
 - The Rise of Digital Reading Acids Bases Section 1 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Acids Bases Section 1 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acids Bases Section 1 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acids Bases Section 1 Answers
 - Personalized Recommendations
 - Acids Bases Section 1 Answers User Reviews and Ratings
 - Acids Bases Section 1 Answers and Bestseller Lists

5. Accessing Acids Bases Section 1 Answers Free and Paid eBooks
 - Acids Bases Section 1 Answers Public Domain eBooks
 - Acids Bases Section 1 Answers eBook Subscription Services
 - Acids Bases Section 1 Answers Budget-Friendly Options
6. Navigating Acids Bases Section 1 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Acids Bases Section 1 Answers Compatibility with Devices
 - Acids Bases Section 1 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acids Bases Section 1 Answers
 - Highlighting and Note-Taking Acids Bases Section 1 Answers
 - Interactive Elements Acids Bases Section 1 Answers
8. Staying Engaged with Acids Bases Section 1 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acids Bases Section 1 Answers
9. Balancing eBooks and Physical Books Acids Bases Section 1 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acids Bases Section 1 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Acids Bases Section 1 Answers
 - Setting Reading Goals Acids Bases Section 1 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acids Bases Section 1 Answers
 - Fact-Checking eBook Content of Acids Bases Section 1 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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