

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

Chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a homogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution.

1. Define the term solvent and provide an example.

2. Define solubility, and how is it measured?

3. Distinguish between concentrated and dilute.

4. List the properties of bases and acids.

5. Explain the significance of the pH scale.

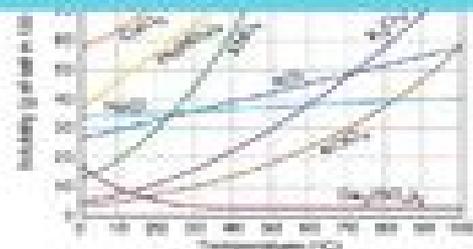
6. How does the concentration of an acid affect its pH?

7. Explain the process of neutralization.

8. Explain the difference between strong acids and weak acids.

9. Explain the difference between strong bases and weak bases.

10. Explain the difference between a strong acid and a weak acid.



11. Describe how you would use the graph to determine the amount of NaCl that can be dissolved in 100 grams of water at 60°C .

12. Which salt has the greatest solubility at 80°C ?



ANSWER KEY INCLUDED

Acids And Bases Study Guide Key

R Barnett



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