

Name: \_\_\_\_\_

Date: \_\_\_\_\_

### Chapter 6 Study Guide: Periodicity

1. Describe the historical development of the periodic table (Newland, Meyer, Mendeleev, Mosely) Describe the contributions made by each of the scientist.

**Newland** – First to order the elements in a periodic table –developed the law of octaves

**Meyers** – Ordered elements by atomic mass but is given little credit for the current periodic table

**Mendeleev** – ordered elements by atomic mass and is given the most credit for the current periodic table because he left holes in his table for undiscovered elements and described their properties

**Mosely** – ordered elements by atomic number (modern periodic table)

2. Fill in the blanks below:

The elements in the periodic table are arranged in order of increasing Atomic Number. The vertical columns are called Groups or Families. The horizontal rows are called Periods. The elements within the same Family have the same chemical properties due to the number of their Valence Electrons. Metals are located on the Left side of the periodic table. The nonmetals are located on the Right side of the periodic table. The metalloids are located on the Stair step between the metals and nonmetals.

3. Describe the general properties of the following groups of elements on the periodic table:

a. Alkali Metals

**1 valence electrons, metals, highly reactive, s-block**

b. Alkaline Earth metals

**2 valence electrons, metals, reactive, s-block**

c. Halogens

**7 valence electrons, high reactive, nonmetals, p-block**

d. Noble Gases

**8 noble electrons, nonreactive, nonmetals, all gases, p-block**

e. Transition Metals

**Variable valence electrons, metals, malleable, ductile, metals, conductive**

f. Inner Transition Metals

**Variable valence electrons, metals, f-block**

# Chapter 6 Periodic Table Answers

**M Tight**



## Chapter 6 Periodic Table Answers:

Cambridge Checkpoint Lower Secondary Science Revision Guide for the Secondary 1 Test 2nd edition Rosemary Feasey, Andrea Mapplebeck, David Bailey, 2023-04-06 Achieve maximum potential using step by step guidance that helps to practise skills learned and improve exam technique Build confidence with practical study tips and effective revision strategies Reinforce understanding with clear explanations of every topic covered in the Cambridge Lower Secondary Checkpoint curriculum frameworks Strengthen and test knowledge with a range of questions and worked examples Test questions worked solutions and answers are available free online at [www.hoddereducation.co.uk/cambridgeextras](http://www.hoddereducation.co.uk/cambridgeextras) This resource has not been through the Cambridge International endorsement process

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Applications takes a technical approach to address these issues using green design and analysis This book provides an overview of the latest developments in environmental chemistry and sustainable materials written by experts in their respective research areas This interdisciplinary volume offers research with the aim to minimize environmental impacts across all lifecycle phases in the design and engineering of products processes and systems as just one possible approach to addressing the larger issue of sustainability that includes environmental economic and social aspects Materials Principles and Practice Charles Newey, Graham Weaver, 2013-10-22 Materials Principles and Practice deals with materials science in the technological context of making and using materials Topics covered include the nature of materials such as crystals an atomic view of solids temperature effects on materials and the mechanical and chemical properties of materials This book is comprised of seven chapters and begins with an overview of the properties of different kinds of material the ways in which materials can be shaped and the uses to which they can be put The next chapter describes the state of matter as a balance between the tendencies of atoms to stick together by chemical bonding or rattle apart by thermal agitation paying particular attention to ionic bonds and ionic crystals the structure and properties of polymers and transition metals The reader is also introduced to how the structure of materials especially microstructure can be manipulated to give desired properties via thermal mechanical and chemical agents of change This text concludes by describing the chemistry of processing and service of various materials Exercises and self assessment questions with answers are given at the end of each chapter together with a set of objectives This monograph will be a valuable resource for students of materials science and the physical sciences

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