

## Short Answer Question

**Q.1 Define concentration process, is it used in metallurgy of copper?**

**Ans.** The process of removal of gangue from the ore is technically known as concentration and the purified ore is called the concentrate. Yes, concentration process used in metallurgy of copper.

**Q.2 Why a small amount of coke is required in the smelting process?**

**Ans.** Because smelting is carried out in blast furnace. The process in blast furnace is highly exothermic process. Therefore a small amount of coke is required in this process.

**Q.3 Why lime is added in the smelting process?**

**Ans.** Lime is added to remove excess of  $\text{SiO}_2$ . Lime reacts with sand to form slag.



**Q.4 How slag and matte are removed from the blast furnace?**

**Ans.** Slag being lighter rise to the top and is removed from the upper hole of the blast furnace and matte is withdrawn from the lower hole of the blast furnace. It contains about 45% of copper.

**Q.5 What is the difference between slag and matte?**

**Ans.**

Slag	Matte
When flux combine with gangue it will form slag which being lighted in weight and floats on the molten metal	In blast furnace cuprous sulphide and ferrous sulphide form a mixture ( $\text{Cu}_2\text{S}$ , $\text{FeS}$ ). This molten mixture is called matte.

**Q.6 Mention the chemical reaction for the formation of metallic copper in the bessemerization process.**

**Ans.** Following, chemical reactions for the formation of metallic copper in the bessemerization process



# Chapter 8 Chemistry Answers

**L Towne**



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