

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ **lattice energy** _____ a. ionic bond
 _____ **bond energy** _____ b. covalent bond
 _____ **enthalpy of vaporization** _____ c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ **ionic** _____ a. H—F _____ **nonpolar covalent** _____ d. H—H
 _____ **ionic** _____ b. Na—Cl _____ **polar covalent** _____ e. H—C
 _____ **polar covalent** _____ c. H—O _____ **polar covalent** _____ f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Hydrogen bonds are generally intermolecular, while ionic and covalent bonds occur between ions or atoms respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond. Increased polarity in H₂O bonds means a stronger intermolecular attraction, making water a liquid at room temperature. Hydrogen bonding exists between water molecules, but not between hydrogen sulfide molecules.

Chapter 8 Covalent Bonding Section 81 Molecular Compounds Answers

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Hren, Christopher R. Hren**



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Chemistry Richard Post, Chad Snyder, Clifford C. Houk, 2020-09-16 A practical complete and easy to use guide for understanding major chemistry concepts and terms Master the fundamentals of chemistry with this fast and easy guide Chemistry is a fundamental science that touches all other sciences including biology physics electronics environmental studies astronomy and more Thousands of students have successfully used the previous editions of Chemistry Concepts and Problems A Self Teaching Guide to learn chemistry either independently as a refresher or in parallel with a college chemistry course This newly revised edition includes updates and additions to improve your success in learning chemistry This book uses an interactive self teaching method including frequent questions and study problems increasing both the speed of learning and retention Monitor your progress with self tests and master chemistry quickly This revised Third Edition provides a fresh step by step approach to learning that requires no prerequisites lets you work at your own pace and reinforces what you learn ensuring lifelong mastery Master the science of basic chemistry with this innovative self paced study guide Teach yourself chemistry refresh your knowledge in preparation for medical studies or other coursework or enhance your college chemistry course Use self study features including review questions and quizzes to ensure that you re really learning the material Prepare for a career in the sciences medicine or engineering with the core content in this user friendly guide Authored by expert postsecondary educators this unique book gently leads students to deeper levels and concepts with practice critical thinking problem solving and self assessment at every stage

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Advanced Inorganic Chemistry - Volume I Satya Prakash et al., 2000-10 Advanced Inorganic Chemistry Volume I is a

concise book on basic concepts of inorganic chemistry It acquaints the students with the basic principles of chemistry and further dwells into the chemistry of main group elements and their compounds It primarily caters to the undergraduate courses Pass and Honours offered in Indian universities Chemistry Steven S. Zumdahl,1989 **Chemistry Workbook For Dummies with Online Practice** Chris Hren,Peter J. Mikulecky,2017-04-17 Take the confusion out of chemistry with hundreds of practice problems Chemistry Workbook For Dummies is your ultimate companion for introductory chemistry at the high school or college level Packed with hundreds of practice problems this workbook gives you the practice you need to internalize the essential concepts that form the foundations of chemistry From matter and molecules to moles and measurements these problems cover the full spectrum of topics you ll see in class and each section includes key concept review and full explanations for every problem to quickly get you on the right track This new third edition includes access to an online test bank where you ll find bonus chapter quizzes to help you test your understanding and pinpoint areas in need of review Whether you re preparing for an exam or seeking a start to finish study aid this workbook is your ticket to acing basic chemistry Chemistry problems can look intimidating it s a whole new language with different rules new symbols and complex concepts The good news is that practice makes perfect and this book provides plenty of it with easy to understand coaching every step of the way Delve deep into the parts of the periodic table Get comfortable with units scientific notation and chemical equations Work with states phases energy and charges Master nomenclature acids bases titrations redox reactions and more Understanding introductory chemistry is critical for your success in all science classes to follow keeping up with the material now makes life much easier down the education road Chemistry Workbook For Dummies gives you the practice you need to succeed **Fundamentals of General, Organic, and Biological Chemistry** John R. Holum,1982 User friendly this comprehensive and up to date edition is organized according to the principle that life in both health and disease has a molecular basis Carefully worked example exercises enhanced by Analysis and Check sections enable students to examine and think about a problem not just answer it Includes numerous environmental applications **Crystallography Reports** ,1999-07 Current Russian research in crystallography *Biological & Agricultural Index* ,1997 *How to Prepare for College Board Achievement Tests* Joseph A. Mascetta,1990 **Science Citation Index** ,1994 Vols for 1964 have guides and journal lists **Monthly Index of Russian Accessions** ,1960 **British Chemical and Physiological Abstracts** ,1950 *Government Reports Announcements & Index* ,1988 **British Abstracts** ,1951 **Applied Science & Technology Index** ,2000 **Enamel Bibliography and Abstracts** ,1953

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 - Chapter 2: Essential Elements of Chapter 8 Covalent Bonding Section 81 Molecular Compounds Answers
 - Chapter 3: Chapter 8 Covalent Bonding Section 81 Molecular Compounds Answers in Everyday Life
 - Chapter 4: Chapter 8 Covalent Bonding Section 81 Molecular Compounds Answers in Specific Contexts
 - Chapter 5: Conclusion
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- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 8 Covalent Bonding Section 81 Molecular Compounds Answers.

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