

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ **lattice energy** _____ a. ionic bond
 _____ **bond energy** _____ b. covalent bond
 _____ **enthalpy of vaporization** _____ c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ **ionic** _____ a. H—F _____ **nonpolar covalent** _____ d. H—H
 _____ **ionic** _____ b. Na—Cl _____ **polar covalent** _____ e. H—C
 _____ **polar covalent** _____ c. H—O _____ **polar covalent** _____ f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Hydrogen bonds are generally intermolecular, while ionic and covalent bonds occur between ions or atoms respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond. Increased polarity in H₂O bonds means a stronger intermolecular attraction, making water a liquid at room temperature. Hydrogen bonding exists between water molecules, but not between hydrogen sulfide molecules.

Chapter 8 Covalent Bonding Test B Answers

Tao Wei



Chapter 8 Covalent Bonding Test B Answers:

Enjoying the Melody of Phrase: An Emotional Symphony within **Chapter 8 Covalent Bonding Test B Answers**

In a world consumed by displays and the ceaseless chatter of quick transmission, the melodic beauty and psychological symphony developed by the prepared word frequently diminish into the background, eclipsed by the persistent sound and disruptions that permeate our lives. But, located within the pages of **Chapter 8 Covalent Bonding Test B Answers** a marvelous fictional treasure full of natural emotions, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, that captivating masterpiece conducts readers on a psychological journey, skillfully unraveling the hidden tunes and profound influence resonating within each cautiously crafted phrase. Within the depths of this emotional analysis, we can explore the book's key harmonies, analyze their enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers' souls.

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