

## Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O<sub>2</sub> 1.

How many moles of oxygen are produced from twelve moles of Potassium chlorate?

21 mol O<sub>2</sub> 2.

How many moles of oxygen are produced from 14 moles of Potassium chlorate?

45 mol H<sub>2</sub>O 3.

How many moles of water are produced from 22.5 moles of oxygen?

20 mol HCl 4.

Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

21 mol Cl<sub>2</sub> 5.

Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



# Chapter 8 Stoichiometry Review Sheet Answers

**Pamela Mills, Amina El-Ashmawy**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author names.

## Chapter 8 Stoichiometry Review Sheet Answers:

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