

Solutions

$$\Delta S_{mix} = +ve, \Delta G_{mix} = -ve \quad (\text{Always})$$

21. Non-ideal solution :- Do not obey Raoult's law.

A-A or B-B \neq A-B (FOA) + Properties :- $\Delta H_{mix} \neq 0, \Delta V_{mix} \neq 0, P_3 \neq P_A^0 X_A + P_B^0 X_B$

→ +ve deviation from Raoult's law :- A-A + B-B $>$ A-B (FOA)
 Properties :- $P_3 > P_A^0 X_A + P_B^0 X_B$ (Ex. Acetone + Alcohol (ethanol); Benzene + methanol)
 (b) $\Delta H_{mix} = +ve$ (Endothermic rxn, heat absorb) (c) $\Delta V_{mix} = +ve$ Acetone + Benzene

→ -ve deviation from Raoult's law :- A-A + B-B $<$ A-B (FOA)
 Properties :- $P_3 < P_A^0 X_A + P_B^0 X_B$ (Ex. Strong acid + water, Acetone + $CHCl_3$, Phenol + aniline, $CHX_3 + (O, N)$ containing compounds)

22. Azeotropes :- A mixture of two or more components which boils like pure chemical compound and distills over completely at the same temp without change in chemical composition.

- (i) Minimum boiling azeotropes → B.P. is less than B.P. of either component
- (ii) Maximum boiling azeotropes → B.P. is more than B.P. of either component

Ex. +ve deviation from Raoult's law Ex. -ve deviation from Raoult's law

Note :- $Y_A = P_A^0 X_A - P_A^0 X_A - P_A^0 X_A$...

Chemistry Note For Iit Jee

Jicheng Xie



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