

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) *79* copper (Cu) *29*

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) *35.5* calcium (Ca) *40.1*

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
<i>Lead</i>	<i>82</i>	<i>207</i>	<i>82</i>	<i>125</i>	<i>82</i>
<i>Argon</i>	<i>18</i>	<i>38</i>	<i>18</i>	<i>20</i>	<i>18</i>
<i>Barium</i>	<i>56</i>	<i>137</i>	<i>56</i>	<i>81</i>	<i>56</i>
<i>Tin</i>	<i>50</i>	<i>112</i>	<i>50</i>	<i>62</i>	<i>50</i>

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

Chemistry Chapter9 Review Answers

The Princeton Review



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