

Chapter Test II

A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

Column I

- _____ 1. An oxidation-reduction reaction in which an element is oxidized and another element is reduced.
- _____ 2. An oxidation-reduction reaction in which a substance is oxidized and the same substance is reduced.
- _____ 3. The reduction of an oxidizing agent.
- _____ 4. The sum of the oxidation numbers of all atoms in a molecule.
- _____ 5. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
- _____ 6. An oxidation-reduction reaction in which a substance is oxidized and another substance is oxidized.

Column II

- a. oxidation
- b. limiting reagent
- c. oxidation state
- d. reduction
- e. oxidation-reduction reaction
- f. oxidation-reduction reaction

B. Multiple Choice

Choose the best answer for each question.

- _____ 1. The oxidation number of sulfur in H_2SO_4 is
 - a. +2
 - b. +4
 - c. +6
 - d. +8
- _____ 2. Which of the following is an oxidation-reduction reaction?
 - a. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - b. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}_2$
 - c. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$
 - d. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$
- _____ 3. The sum of the oxidation numbers of all atoms in a molecule is
 - a. 0
 - b. 1
 - c. 2
 - d. 3

Chemistry Chapter 12 Test B Stoichiometry

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