

Chemical Reactions

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolysis Reaction

1. Write the chemical formula following compounds:

a) hydrogen gas: H_2 b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $Mg_3(PO_4)_2$ d) sodium nitrate: $NaNO_3$

2. Write the correct formula following compounds:

a) $Ca(NO_3)_2$: $Ca(NO_3)_2$ b) $FeCl_3$: $FeCl_3$
 c) $CaSO_4$: $CaSO_4$ d) NaF : NaF

Chemical Equations: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that you are going to describe what happens during:

Chemical Equations react with **physical** and **chemical** **properties** and **physical** **properties**

There are also some characteristics that you are going to see and the reactions themselves. Here are a few examples:

Critical Thinking Questions

1. Write the following reaction, with the correct gas laws with it:



Chemical Equations react with **physical** and **chemical** **properties** and **physical** **properties**

Chemquest 34 Answers

**Douglas Buchanan, James Melrose, A. R.
Ross**

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Training and Education for Online Institute of Information Scientists,1989
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