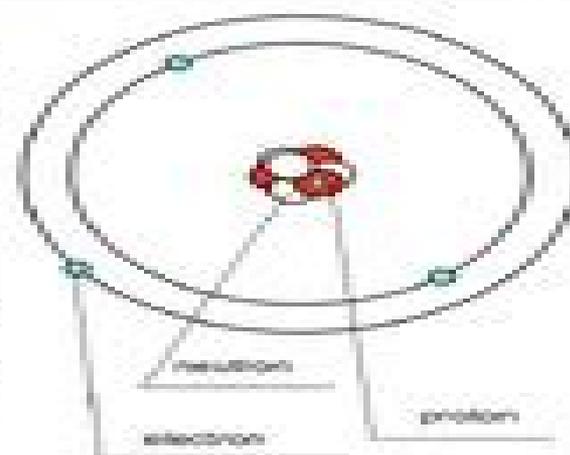


Answer Key

1. **PROTONS** are found in the center of the atom in a region called the **NUCLEUS**. (slide 7)
2. It is the number of **protons** that determine what **ELEMENT** an atom will be. (slide 8)
3. The number of **protons** is called the **ATOMIC NUMBER**. (like an ID Number) (slide 9)
4. The **atomic number** is a whole number found with the **SYMBOL** of the element. (slide 10)
5. **Protons** have a mass of **1 amu** and are one of the heavier pieces of an atom. (slide 11)
6. **Protons** have a **+1** charge. (slide 12)
7. **NEUTRONS** are the other type of particle found in the center of the atom (nucleus). (slide 13)
8. **Neutrons** have **ZERO** charge and have a mass of **1 amu**. (slide 14-15)
9. **ELECTRONS** are the smallest of the subatomic particles, found outside the nucleus. (slide 16)
10. **Electrons** have virtually no significant **MASS**. (slide 16)
11. **Electrons** have a **-1** charge. (slide 17)
12. **Electrons** can **MOVE** between atoms or be **SHARED** between atoms. (slide 17)
13. If an atom is **NEUTRAL**, it will have the same number of electrons as protons. (slide 20)
14. **Electrons** are involved in **all CHEMICAL REACTIONS**. (slide 21)

PARTICLE	MASS	CHARGE	LOCATION	ROLE
PROTON	1	+1	Nucleus	Identity of Element
NEUTRON	1	0	Nucleus	Holds Nucleus Together
ELECTRON	0	-1	Outside Nucleus	Reactivity



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Lex Sharp



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Explains what an atom is and why it is important and describes the particles that make up atoms *Elementary Atomic Structure* Gordon Kemble Woodgate,1998 **Chemistry for Kids, the Atomic Structure** Lex Sharp,2018-03-24 This book is meant for students grades 6 and higher It addresses concepts that are critical to the understanding of the inner structure of the atom which lays the foundation for appreciating how elements behave in chemical reactions The text has been integrated with colorful images that help engage the visual with the conceptual No prerequisites are necessary The book is suitable for the beginner who has not studied chemistry before and for the student who wants to clarify questions for review purposes A couple of mathematical chapters were also infused into the book to support younger learners with the necessary mathematical background to succeed To get the most from this book students should summarize each chapter and rehearse the concepts Attempting to work independently through all problems at the end of the book is beneficial only before reading the solutions When solving problems one can be sure the chemical principles presented were well understood Restricting oneself to only reading is incomplete learning The series walks learners gradually and conceptually through the chemical and particle physics theories that are meant to jumpstart a more profound scientific understanding Topics covered in this volume Subatomic Particle Atomic Mass Period Table of Elements Atomic Number Isotope and Ion Notations Atomic Weight The Mole Molar Mass Exponents The Rule of Three Practice problems and easy to follow solutions

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1. Understanding the eBook Atomic Structure Guided Notes Answers Boypic
 - The Rise of Digital Reading Atomic Structure Guided Notes Answers Boypic
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Guided Notes Answers Boypic
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Guided Notes Answers Boypic
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Guided Notes Answers Boypic
 - Personalized Recommendations
 - Atomic Structure Guided Notes Answers Boypic User Reviews and Ratings
 - Atomic Structure Guided Notes Answers Boypic and Bestseller Lists

5. Accessing Atomic Structure Guided Notes Answers Boypic Free and Paid eBooks
 - Atomic Structure Guided Notes Answers Boypic Public Domain eBooks
 - Atomic Structure Guided Notes Answers Boypic eBook Subscription Services
 - Atomic Structure Guided Notes Answers Boypic Budget-Friendly Options
6. Navigating Atomic Structure Guided Notes Answers Boypic eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Guided Notes Answers Boypic Compatibility with Devices
 - Atomic Structure Guided Notes Answers Boypic Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Guided Notes Answers Boypic
 - Highlighting and Note-Taking Atomic Structure Guided Notes Answers Boypic
 - Interactive Elements Atomic Structure Guided Notes Answers Boypic
8. Staying Engaged with Atomic Structure Guided Notes Answers Boypic
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Guided Notes Answers Boypic
9. Balancing eBooks and Physical Books Atomic Structure Guided Notes Answers Boypic
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Guided Notes Answers Boypic
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Guided Notes Answers Boypic
 - Setting Reading Goals Atomic Structure Guided Notes Answers Boypic
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure Guided Notes Answers Boypic
 - Fact-Checking eBook Content of Atomic Structure Guided Notes Answers Boypic
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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