

Fuel Cells Harvest Chemical Energy

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The fuel cell converts chemical energy from a hydrocarbon fuel (such as hydrogen, butanol, and other fuels) directly to electrical energy through a complex process called electrochemistry. In this process, hydrocarbons react with oxygen and water to produce and the energy released is used to drive a motor or generator. Hydrogen, O₂ and water are the main components of the fuel cell. Hydrogen is used to drive a motor or generator, and the energy released is used to drive a motor or generator. The fuel cell is a device that converts chemical energy from a hydrocarbon fuel (such as hydrogen, butanol, and other fuels) directly to electrical energy through a complex process called electrochemistry.

Applying Your Knowledge

Answer the questions below and complete additional responses to the questions below.

1. What are the main components of a fuel cell? How do they work together to produce electricity? (20 points)

2. How does a fuel cell differ from a battery? (10 points)

3. What are the advantages and disadvantages of fuel cells? (10 points)

4. How do fuel cells compare to other energy sources? (10 points)

5. How do fuel cells compare to other energy storage devices? (10 points)

6. How do fuel cells compare to other energy conversion devices? (10 points)

7. How do fuel cells compare to other energy conversion devices? (10 points)

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Donald W. Buck II



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