

IONIC & COVALENT BONDS
 Honors Chemistry Lab
 Adapted from SAS Curriculum Pathways Lab

1. Refer to these websites (or choose your own web resources) to collect the data required to complete the chart. When filling out the chart, try your best to keep the same units across a row. Use an online unit converter if necessary.

[PubChem](#)
[American Elements](#)
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2. Use the data in the chart to answer the analysis questions.

Name	Barium oxide	Bromine (in diatomic form)	Carbon tetrachloride	Glucose
Formula	BaO	Br ₂		
Structure (show correct shape)				
Melting point (°C)				
Boiling point (°C)				N/A
Physical state at room temperature				
Solubility in water (indicate unit)				
Electronegativity of each element	Ba = O =	Br = Br =	C = Cl =	O = H =
EN Difference				
Type of Bond (ionic, nonpolar covalent, polar covalent)				

Assignment On Ionic And Covalent Compounds Sss Chemistry

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Synthetic Organic Chemicals, 1927 CRC Handbook of Chemistry and Physics Robert C. Weast, 1975

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