

Molecule Polarity Simulation

Visit the [Molecule Polarity simulator](#).

1. Select the "Real Molecules" simulation option (click on the "Real Molecules Only in Java" link; you will then be prompted to download the simulation). Once you download the simulation to your device, open the simulation and begin the activity.

Set 1: HF vs. H₂

For each molecule be sure to have the "Atom Labels" and "Atom Electronegativities" button selected. Use the "Bond Dipoles" and "Molecular Dipoles" buttons to determine if these dipoles are present when instructed below.

- Use the simulation to determine if bond dipoles are present on each molecule.
- Use the simulation to determine if a molecular dipole is present on each molecule.
- Use the simulation "Electrostatic Potential" option to map the electron density on each molecule.

Molecule	Electronegativity Difference between atoms	Bond Dipoles Present? (yes or no)	Molecular Dipole Present? (yes or no)	Sketch/describe the Electrostatic Potential
H ₂	$2.1 - 2.1 = 0$	no	no	
HF	$4.0 - 2.1 = 1.9$	yes	yes	

Question 1: Using what you learned from Set 1, predict what properties of the molecule determine if a molecular dipole is present or not.

Molecules with non-polar bonds will be non-polar. Molecules with polar bonds might be polar.

Set 2: Repeat the data collection carried out in Set 1 for each set molecules below. Be sure to have the "Atom Labels" "Atom Electronegativities" button selected. Use the "Bond Dipoles" and "Molecular Dipoles" buttons to determine if these dipoles are present when instructed below.

- Use the simulation to determine if bond dipoles are present on each molecule.
- Use the simulation to determine if a molecular dipole is present on each molecule.
- Use the simulation "Electrostatic Potential" option to map the electron density on each molecule.

Phet Simulation Molecule Polarity Lab Answers

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terbantahkan karena sudah ditemukan senyawa gas mulia dan gas langka yang terbantahkan karena argon ternyata mengisi sekitar 0.94 volume atmosfer bumi helium adalah gas mulia yang pertama ditemukan konfigurasi elektron

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apr 16 2017 mempunyai titik didih dan titik leleh yang sangat rendah jari jari atom gas mulia sangat kecil mempunyai elektron valensi 8 khusus untuk helium elektron valensinya 2 sudah stabil sehingga sukar membentuk senyawa dengan unsur lain di alam tidak pernah di temukan senyawa gas mulia gas mulia terdapat sebagai molekul monoatomik

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fgas mulia dan halogen sejarah gas mulia golongan gas mulia atau golongan viii a adalah unsure unsur yang memiliki delapan elektron valensi dengan konfigurasi elektronik terluar $ns^2 np^6$ unsur unsur tersebut adalah helium he neon

gas mulia pengertian sifat kegunaan pembuatan konfigurasi - Jun 28 2023

oct 6 2023 gas mulia ditemukan di dekat golongan halogen dalam tabel periodik karena unsur gas mulia memiliki konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas gas ini dikenal dengan gas inert

apa itu halogen dalam kimia berikut definisi unsur sifat dan - Nov 21 2022

may 18 2022 dikutip dari buku kimia unsur golongan gas mulia halogen alkali dan alkali tanah yang disusun oleh tantri ambarsari 2020 halogen adalah senyawa yang unsur unsurnya dapat bereaksi dengan logam untuk membentuk garam reaktivitas unsur halogen sangat besar sehingga unsur unsurnya sulit ditemukan di alam