

Volumetric Analysis-I

Prepare 0.05M solution of Ferrous ammonium sulphate (Mohr's salt). Using this solution find out the molarity and strength of the given KMnO_4 solution.

Aim:

To Prepare 0.05M solution of ferrous ammonium sulphate (Mohr's salt). Using this solution find out the molarity and strength of the given KMnO_4 solution.

Chemical Equations:



Ionic equations:



Result:

- Molarity of the KMnO_4 solution =
- Strength of the KMnO_4 solution =

Short Procedure:

S.No	TITRATION	
1	Burette solution	Unknown solution of KMnO_4
2	Pipette solution	20mL of standard solution of Mohr's salt
3	Add a solution	20mL of 2M of H_2SO_4
4	Temperature	Room temperature
5	Indicator	Self-indicator (KMnO_4)
6	End point	Appearance of permanent pale pink colour
Equivalent mass of Mohr's salt = 392		

Observation:

- Weight of watch glass = g
Weight of watch glass + Mohr's salt = g
Weight of Mohr's salt = g
Volumetric of Mohr's salt solution prepared = 250ml

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