

THERMODYNAMICS SAMPLE PROBLEM

1. Find the work possess for a helium gas at 20°C.

- A. 609 kJ/kg B. 168 kJ/kg C. 229 kJ/kg D. 339 kJ/kg

2. Two kilogram of gas is confined in a 1 m³ tank at 200 kPa and 88°C. What type of gas is in the tank?

- A. Helium B. Ethane C. Methane D. Ethene

3. Find the enthalpy of Helium if its internal energy is 200 kJ/kg.

- A. 144 kJ/kg B. 223.42 kJ/kg
C. 333.42 kJ/kg D. 168 kJ/kg

4. Compute the mass of a 2 m³ propane at 280 kPa and 40°C.

- A. 6.47 kg B. 5.1 kg C. 10.20 kg D. 9.47 kg

5. A thermal power plant has the heat rate of 11,363 Btu/kW-hr. Find the thermal efficiency of the plant.

- A. 28% B. 30% C. 34% D. 40%

6. Find the mass of carbon dioxide having a pressure of 20 psia at 200°F with 10 ft³ volume.

- A. 1.04 lbs B. 1.14 lbs C. 1.24 lbs D. 1.34 lbs

7. Two kilogram of air in a rigid tank changes its temperature from 32°C to 150°C. Find the work done during the process.

- A. 236 B. 170 C. 195 D. 0

8. Determine the atmospheric pressure at a location where barometric reading is 740 mm Hg and gravitational acceleration is $g = 9.7 \text{ m/s}^2$. Assume the temperature of mercury to be 10°C, at which the density is 13,570 kg/m³.

- A. 99.45 kPa B. 97.41 kPa
C. 95.44 kPa D. 98.66 kPa

9. A perfect gas has a value of $R = 58.8 \text{ ft}\cdot\text{lb}/\text{lb}\cdot^\circ\text{R}$ and $k = 1.26$. If 20 Btu are added to 10 lbs of this gas at constant volume when initial temperature is 50°F, find the final temperature.

- A. 97°F B. 104°F C. 154°F D. 185°F

10. A tank contains 90 ft³ of air at a pressure of 350 psig; if the air is cooled until its pressure and temperature decreases at 200 psig and 70°F respectively, what is the decrease in internal energy?

- A. 6232.09 Btu B. -5552 Btu

C. 5552 Btu

D. -6232.09 Btu

11. A spherical balloon with a diameter of 6 m is filled with helium at 20°C and 200 kPa. Determine the mole number.

- A. 9.28 kmol B. 10.28 kmol
C. 11.28 kmol D. 13.28 kmol

12. The air in an automobile tire with a volume of 0.53 ft³ is at 90°F and 20 psig. Determine the amount of air that must be added to raise the pressure to the recommended value of 30 psig. Assume the atmospheric pressure to be 14.7 psia and the temperature and the volume to remain constant.

- A. 0.026 lb B. 0.046 lb
C. 0.066 lb D. 0.086 lb

13. An elastic sphere containing gas at 120 kPa has a diameter of 1.0 m. H eating the sphere causes it to expand to a diameter of 1.3 m. During the process the pressure is proportional to the sphere diameter. Calculate the work done by the gas in kJ.

- A. 41.8 B. 50.6 C. 87.5 D. 35.4

14. A rigid tank contains 5 kg of an ideal gas at 4 atm and 40°C. Now a valve is opened, and half of mass of the gas is allowed to escape. If the final pressure in the tank is 1.5 atm, the final temperature in the tank is?

- A. -38°C B. -30°C C. 40°C D. 53°C

15. A pressure cooker operates by cooking food at a higher pressure and temperature possible at atmospheric conditions. Steam is contained in the sealed pot, with vent hole in the middle of the cover, allowing steam to escape. The pressure is regulated by covering the vent hole with a small weight, which is displaced slightly by escaping steam. Atmospheric pressure is 100 kPa, the vent hole area is 7-mm², and the pressure inside should be 250 kPa. What is the mass of the weight?

- A. 0.107 kg B. 1.05 kg
C. 1.75 kg D. 0.1783 kg

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