

K_w CALCULATION

$$K_w = [H^+][OH^-]$$



$$\Rightarrow [H^+] = \frac{K_w}{[OH^-]} = \frac{1 \times 10^{-14}}{0.46}$$

$$pH = -\log[H^+]$$

$$= -\log(2.17391 \times 10^{-14})$$

$$= 13.66$$

pH to 2 d.p.!

A-LEVEL CHEMISTRY

Calculations For A Level Chemistry

JE Gale



Calculations For A Level Chemistry:

Calculations For A Level Chemistry Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Calculations For A Level Chemistry**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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