

Chapter 4 Chemical Bonds

**Word Wise**

Use the terms from the following list to fit each of the clues given below.

molecular	covalent	polyatomic
crystals	anion	polar
ionic	alloy	metallic
formula	bond	cation

**Clues**

**Vocabulary Terms**

- A type of bond that holds cations and anions together \_\_\_\_\_
- A type of bond in which two atoms share a pair of valence electrons \_\_\_\_\_
- A neutral group of atoms that are joined together by one or more covalent bonds \_\_\_\_\_
- A term describing a covalent bond in which electrons are not shared equally \_\_\_\_\_
- An ion that contains a covalently bonded group of atoms \_\_\_\_\_
- An ion with a negative charge \_\_\_\_\_
- An ion with a positive charge \_\_\_\_\_
- A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound \_\_\_\_\_
- Solids whose particles are arranged in a lattice structure \_\_\_\_\_
- A mixture of two or more elements, at least one of which is a metal \_\_\_\_\_
- A type of bond that exists between a metal cation and the shared electrons that surround it \_\_\_\_\_
- The force that holds atoms or ions together \_\_\_\_\_

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# Chapter 6 Chemical Bonds Wordwise Answer Key

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