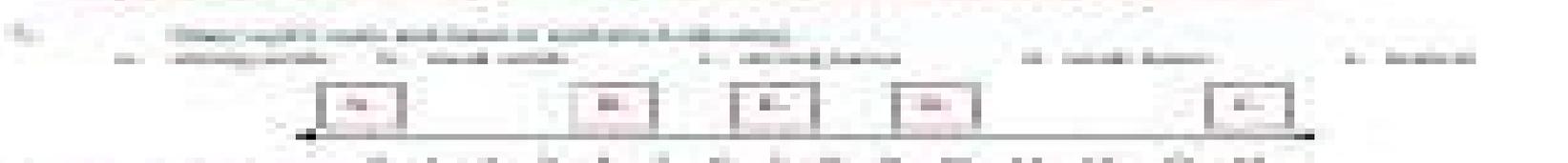


1. Zeichnen Sie die folgenden Komplexe, geben Sie die Oxidationszahl des Zentralatoms an!

- a) $[PtCl_4]^{2-}$ → Oxidationszahl Zentralatom: +2
- b) $[PtCl_6]^{2-}$ → Oxidationszahl Zentralatom: +4
- c) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- d) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- e) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- f) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- g) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- h) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4

- a) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- b) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- c) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- d) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- e) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- f) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- g) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- h) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4

Zeichnen Sie die folgenden Komplexe, geben Sie die Oxidationszahl des Zentralatoms an!



Zeichnen Sie die folgenden Komplexe, geben Sie die Oxidationszahl des Zentralatoms an!



Zeichnen Sie die folgenden Komplexe, geben Sie die Oxidationszahl des Zentralatoms an!



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Zeichnen Sie die folgenden Komplexe, geben Sie die Oxidationszahl des Zentralatoms an!



- a) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- b) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
- c) $[PtCl_2(NH_3)_2]$ → Oxidationszahl Zentralatom: +2
- d) $[PtCl_2(NH_3)_4]^{2+}$ → Oxidationszahl Zentralatom: +4
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Chapter 8 Solutions Acids Bases Assessment Answer Key

Douglas P. Heller, Carl H. Snyder



Chapter 8 Solutions Acids Bases Assessment Answer Key:

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Student Study Guide Fundamentals of Chemistry by David E. Goldberg Margaret G. Kimble, 1994 *Experiments for Living Chemistry* David Ucko, 2012-12-02 Experiments for Living Chemistry provides practical hands on experiments illustrating the concepts substances and techniques that are important to students in the health related sciences Many of these experiments are based on physiological substances to show students how chemical principles apply to the functioning of their own bodies while other experiments use cut outs to help students visualize such complex concepts as bonding and protein synthesis This book is organized into 23 chapters that correspond on a chapter by chapter basis with the Living Chemistry textbook The first five chapters include discussions on matter measurement chemical bonding compounds chemical change gases and respiration The subsequent chapters deal with water solutions acids bases salts hydrocarbons and nuclear and organic

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PCAT Aftab S. Hassan, Williams & Wilkins Review, Betz, 1998 Based on Guidelines from the American Association of Colleges of Pharmacy this volume includes a skills based review of the latest content and a full length practice exam with solutions Readers will acquire the skills and the secrets that will lead to improved success including a 10 point prescription for conquering the exam

Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry Malone, 2000-04-18

A Laboratory Manual of General Chemistry James Henri Walton, Francis Craig Krauskopf, 1920

Introductory Chemistry for Today Spencer L. Seager, Michael R. Slabaugh, 2004 Distinguished by its superior allied health focus and integration of technology Seager and Slabaugh's INTRODUCTORY CHEMISTRY FOR TODAY Fifth Edition continues to lead the market on both fronts through numerous allied health related applications examples boxes and a new Companion Web Site GOB ChemistryNow™ In addition to the many resources found in GOB ChemistryNow this powerful new Web site contains questions modeled after the Nursing School and Allied Health Entrance Exams and NCLEX LPN Certification Exams The authors strive to dispel users inherent fear of chemistry and to instill an appreciation for the role chemistry plays in our daily lives through a rich pedagogical structure and an accessible writing style that provides lucid explanations In addition Seager and Slabaugh's CHEMISTRY FOR TODAY Fifth Edition provides greater support in both problem solving and critical thinking skills By demonstrating how this information will be important to a reader's future career and providing important career information online the authors not only help readers to set goals but also to focus on achieving them

Industrial and Engineering Chemistry, 1923

Olympiad Champs Science Class 10 with 5 Mock Online Olympiad Tests Disha Experts, 2017-07-04 OLYMPIADS Champs Class 10 Science is an attempt to guide and prepare students for Olympiad examinations The book will not only prepare the students for these examinations but will also help in developing a good aptitude and problem solving skills The book covers the complete science portion which is divided into three sections Physics Chemistry and Biology The book provides for each chapter important concepts followed by Multiple Choice Questions Exercises Concepts are summarized in the form of concept map at the end of each chapter Each chapter provides 2 levels of Exercises based on the level of difficulty Each exercise contains Simple MCQs Matching based MCQs statement based MCQs assertion reason based MCQs passage based MCQs and figure picture based MCQs The detailed solutions to the MCQs are provided at the end of each chapter 5 Online mock tests based on the different

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Chemistry for Today William McPherson, William Edwards Henderson, George Winegar Fowler, 1930

Industrial & Engineering Chemistry, 1923

Fundamentals of Chemistry Lotta Jean Bogert, 1924

Study Guide for Chemical Principles, Fourth Edition, by Dickerson, Gray, Darensbourg, and Darensbourg Patricia L. Samuel, 1984

Study Guide for Atoms, Molecules, and Life Michael S. Matta, 1981

Teaching High School Science Alexander Joseph, 1961

Chemical Engineering Catalog, 1920

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004

The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

A Text-book of chemistry and chemical uranalysis for nurses Harold Lindsay Amoss, 1915

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