

## Section 12.1

### Section 12.1

Answer complete answer multiple-choice questions that are provided.

1. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (a) correct, in terms of all the reactants and products.
    - (b) correct, in terms of moles of reactants and products.
    - (c) correct, in terms of work done by work done by the system.
    - (d) correct, in terms of enthalpy change,  $\Delta H$ .
  - b. \_\_\_\_\_
    - (a) the heat capacity of the system.
    - (b) the mass of the system.
    - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
    - (d) the initial enthalpy of the system.
  - c. \_\_\_\_\_
    - (a) correct, in terms of the energy of the system.
    - (b) correct, in terms of the energy of the system.
    - (c) correct, in terms of the energy of the system.
    - (d) correct, in terms of the energy of the system.
  - d. \_\_\_\_\_
    - (a) correct, in terms of the energy of the system.
    - (b) correct, in terms of the energy of the system.
    - (c) correct, in terms of the energy of the system.
    - (d) correct, in terms of the energy of the system.

2. Answer the two questions posed by the following balanced equation:



a. Balance the equation:



b. What is the volume of  $\text{O}_2$  at STP?

\_\_\_\_\_ L

\_\_\_\_\_ g

\_\_\_\_\_ g

# Chapter 9 Stoichiometry Section 1 Answers

**Ying-Ying Zheng**



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