

- (a) What could be metal 'X' and 'Y' ?
- (b) Name the metal salt  $YSO_4$ .
- (c) What type of chemical reaction takes place between X and  $YSO_4$ ? Write the balanced chemical equation.

### LONG TYPE QUESTIONS (5 Marks)

1. White wash was being done at Mukesh's house. Mukesh saw that the painter added quick lime to drum having water. Mukesh touched outer surface of drum, it is unbelievably hot.
- (a) Write the chemical equation for above reaction.
- (b) What type of reaction is it ?
- (c) What is utility of this reaction ?
2. What types of reactions are represented by following :
- (a)  $CaCO_3 \rightarrow CaO + CO_2$
- (b)  $2Ca + O_2 \rightarrow 2CaO$
- (c)  $Pb + CuCl_2 \rightarrow PbCl_2 + Cu$
- (d)  $2FeSO_4 \rightarrow Fe_2O_3 + SO_2 + SO_3$
- (e)  $Na_2SO_4 + BaCl_2 \rightarrow BaCl_2 + 2NaCl$
3. Balance the following equations :
- (a)  $H_2 + O_2 \rightarrow H_2O$
- (b)  $MnO_2 + HCl \rightarrow MnCl_2 + H_2O + Cl_2$
- (c)  $Pb(NO_3)_2 \rightarrow PbO + NO_2 + O_2$

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**Michael Seilmaier**



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