

Nomenclature Review

Name: \_\_\_\_\_

1. Give the charge and name for each of the following *oxyanions*.

	Charge	Name
a. NO <sub>3</sub>	-1	nitrate
b. CO <sub>3</sub>	-2	carbonate
c. C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	-1	acetate
d. ClO <sub>2</sub>	-1	chlorite
e. PO <sub>3</sub>	-3	phosphite
f. ClO <sub>4</sub>	-1	perchlorate
g. SO <sub>4</sub>	-2	sulfate
h. NH <sub>4</sub>	+1	ammonium
i. OH	-1	hydroxide
j. CN	-1	cyanoide
k. BrO <sub>3</sub>	-1	bromate
l. C <sub>2</sub> O <sub>4</sub>	-2	oxalate
m. SO <sub>3</sub>	-2	sulfate
n. MnO <sub>4</sub>	-1	permanganate

2. If A and B represent compounds of an element bonded to a polyatomic ion, give an example of a valid formula for each combination below.

a. AB	b. AB <sub>2</sub>	c. A <sub>2</sub> B	d. AB <sub>3</sub>	e. A <sub>2</sub> B <sub>3</sub>
Al(NO <sub>3</sub> ) <sub>3</sub>	Ca(NO <sub>2</sub> ) <sub>2</sub>	(NH <sub>4</sub> ) <sub>2</sub> S	Al(ClO <sub>4</sub> ) <sub>3</sub>	Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>

3. What is the oxidation number of:

a. Calcium ion?	+2		
b. Sodium ion?	+1		
c. Sulfur ion?	-2		
d. Phosphorus ion?	-3		
e. Carbon ion within Carbon dioxide molecule?	+4	$x + 2(-2) = 0$ $x = 4+$	
f. Phosphorus within the Phosphate polyatomic ion?	+5		$x + 4(-2) = -2$ $x = +6$
g. Chromium within the Chromate polyatomic ion?	+6		
h. Carbon within the Oxalate polyatomic ion?	+3		$x + 4(-2) = -2$ $2x + 4(-2) = -2$ $x = 3+$
i. Zinc ion?	+2		
j. Cadmium ion?	+2		
k. Sulfur within the Hydrogen Sulfite ion (HSO <sub>3</sub> ) <sup>-</sup>	+4	$1 + x - 6 = -1$ $x = 4+$	

4. Give the most common oxidation number/numbers for each of the following groups or families of elements.

a. Alkali Metals	+1
b. Alkaline Earth	+2
c. Noble Gases	0
d. Halogens	-1
e. Group 13	+3
f. Group 15	-3

5. Give the formula for the following hydrocarbons/alcohols:

a. methane	C <sub>1</sub> H <sub>4</sub>
b. butene	C <sub>4</sub> H <sub>8</sub>
c. ethanol	C <sub>2</sub> H <sub>5</sub> OH
d. propyne	C <sub>3</sub> H <sub>4</sub>

# Chemistry Unit 4 Test Review Electron Configuration

**Battelle Memorial Institute**



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web simplifying radicals radicals were introduced in previous tutorial when we discussed real numbers for example  $2\sqrt{5}$  and  $2\sqrt{14}$   $2\sqrt{14} = 2\sqrt{2 \cdot 7} = 2\sqrt{2} \sqrt{7}$  an infinite nonrepeating decimal we are now interested in developing techniques that will aid in simplifying radicals and expressions that contain radicals

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web oct 6 2021 answer  $2yz \sqrt{5x^3y}$  tip to simplify finding an  $n$ th root divide the powers by the index  $a^6 \sqrt{a^3}$  which is  $a^2 \sqrt{a^3}$   $b^6 \sqrt{b^2}$  which is  $b^3 \sqrt{b^6}$   $c^6 \sqrt{c}$  which is  $c^5 \sqrt{c}$  if the index does not divide into the power evenly then we can use the quotient and remainder to simplify

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web aug 12 2022 a radical expression  $\sqrt[m]{a}$  is considered simplified if it has no factors of the form  $m^2$  so to simplify a radical expression we look for any factors in the radicand that are squares definition 6 2 1 for non negative integers  $a$  and  $m$   $a$  is

considered simplified if a has no factors of the form  $m^2$  for example 5 is considered

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web a radical is also in simplest form when the radicand is not a fraction example 1 33 for example has no square factors its factors are 3 11 neither of which is a square number therefore is in its simplest form example 2 extracting the square root 18 has the square factor 9 18 9 2

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web we will simplify this radical expression into the simplest form until no further simplification can be done step 1 find the factors of the number under the radical 486 3 3 3 3 2 step 2 write the number under the radical as a product of its factors as powers of 2 486 3 2 3 2 3 2

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power of the index rewrite the radicand as a product of two factors using that factor step 2 use the product rule to rewrite the radical as the product of two radicals step 3

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