

are composed of single sugar unit. Their formula is multiple of CH_2O . They generally contain carbon number from 3 to 7. They show following characters:

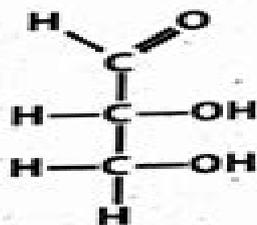
1. They are sweet in taste.
2. They are easily soluble in water. They cannot be hydrolyzed into simple sugars.
3. Except one carbon, all other carbon atoms in a monosaccharide have hydroxyl group (OH). The carbon without OH group has **carbonyl carbon**. It forms aldehyde or ketone group.
4. The sugars with aldehyde groups are called **aldo-sugar** and with the sugar with keto groups are called **keto-sugars**. The example of aldo sugar is glucose and example of keto sugar is fructose.
5. Most of monosaccharides differ in spatial arrangements of their parts around asymmetric carbon. Asymmetric carbon is attached four different types of covalent partners. For example, glucose and galactose differ in arrangement of parts around one asymmetric carbon (carbon no. 3).
6. In aqueous solutions, most monosaccharides form rings.
7. Monosaccharides are major nutrients of cells. They are broken down during cellular respiration to release energy. During respiration many smaller monosaccharides are produced. These smaller monosaccharides like glyceraldehyde, dihydroxy acetone phosphate acts as precursor for the synthesis of many new compounds like amino acids, fatty acids.

Types of Monosaccharides

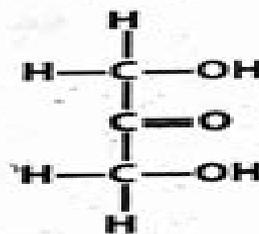
Monosaccharide with 3 to 7 carbon atoms are found in nature. They are called trioses (3C), tetroses (4C), pentoses (5C), hexoses (6C) and heptoses (7C). Their general formula is $(\text{CH}_2\text{O})_n$. There are following types of monosaccharide:

(a) Trioses

They have three carbons. Two trioses are **aldehyde** (glyceraldehydes) and **ketones** (dihydroxyacetone). These are produced as intermediate compounds during respiration and photosynthesis. Thus they have importance in energy transferring system of the cell. These are also used in some metabolic reactions. These trioses provide basic carbon skeleton to different compounds.



**Glyceraldehyde,
an aldotriose**



**Dihydroxyacetone,
a ketotriose**

Biomolecules Structure And Functions 1st Edition

Dev Bukhsh Singh, Timir Tripathi



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