



# ELEMENTARY DIFFERENTIAL EQUATIONS

12<sup>TH</sup> EDITION

# SOLUTIONS

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# Elementary Differential Equations 7th Edition Solution Manual

**JA Banks**



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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1

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web aug 23 2019 these are called polyatomic ions table 2 lists the formulas charges and names of some common polyatomic ions only one of them the ammonium ion is a cation the rest are anions most of them also contain oxygen atoms so sometimes they are referred to as oxyanions

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nitrite perphosphate hypiodite chlorite phosphite percarbonate bromate hyposulfite permanganate carbonite  $\text{co}_3$   $2 \text{mno}_3$   $1 \text{so}_5$   $2 \text{bro}_2$   $1 \text{co}_4$   $2 \text{po}_5$   $3 \text{clo}_1$   $1 \text{io}_2$   $1 \text{po}_4$   $3 \text{no}_2$   $1 \text{so}_4$   $2$

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web polyatomic ions are charged groups of atoms an example is ammonium ion  $\text{nh}_4$  it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is  $\text{nh}_4\text{cl}$  and ammonium sulfide is  $\text{nh}_4$   $2\text{s}$

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web polyatomic ions have defined formulas names and charges that cannot be modified in any way table 3 4 1 3 4 1 lists the ion names and ion formulas of the most common polyatomic ions for example  $\text{no}_3$   $3$  is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion  $\text{no}_3$  contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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names and formulas of some polyatomic ions are found in the below figure and table

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