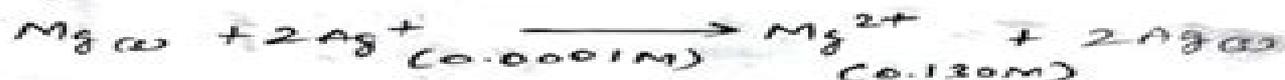


## ELECTROCHEMISTRY NUMERICALS:

### EXAMPLE PROBLEMS:

Prob: 3.1 Represent the cell in which the following reaction takes place



Calculate its  $E_{\text{cell}}$  if  $E_{\text{cell}}^{\circ} = 3.17\text{V}$

Soln:

cell notation:



$$\begin{aligned} E_{\text{cell}} &= E_{\text{cell}}^{\circ} - \frac{0.0591\text{V}}{2} \log \frac{0.130}{(0.0001)^2} \\ &= 3.17\text{V} - 0.02955 \log \frac{0.13}{(10^{-4})^2} \\ &= 3.17 - 0.02955 \log \frac{0.13}{10^{-8}} \\ &= 3.17 - 0.02955 \log (0.13 \times 10^8) \\ &= 3.17 - 0.02955 \log (13 \times 10^6) \\ &= 3.17 - 0.02955 \log 13 + \log 10^6 \\ &= 3.17 - 0.02955 (1.1139 + 6) \\ &= 3.17 - (0.02955 \times 7.1139) \\ \therefore E_{\text{cell}} &= 3.17 - 0.21 = 2.96\text{V} \checkmark \end{aligned}$$

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**Solved Problems in Electrochemistry for Universities and Industry** Dominique L. Piron, 2004      **Electrochemical Methods: Fundamentals and Applications, 3e Student Solutions Manual** Cynthia G. Zoski, John A. Leddy, Allen J. Bard, Larry R. Faulkner, Henry S. White, 2024-11-04 Provides students with solutions to problems in the 3rd edition of the classic textbook *Electrochemical Methods Fundamentals and Applications*. *Electrochemical Methods* is a popular textbook on electrochemistry that takes the reader from the most basic chemical and physical principles through fundamentals of thermodynamics, kinetics, and mass transfer all the way to a thorough treatment of all important experimental methods. Holistically, it offers comprehensive coverage of all important topics in the field. To aid in reader comprehension, exercises are included at the end of each chapter which extend concepts introduced in the text or show how experimental data are reduced to fundamental results. This book provides worked solutions for many of the end of chapter exercises and is a key resource for any student who makes use of the original textbook.

**A Guide to Problems in Modern Electrochemistry 1** Maria E. Gamboa-Aldeco, Robert J. Gale, 2001-09-30 It has been always an incentive for students to find whether his/her efforts to solve exercises give correct results or to find tips for problems that he/she finds more difficult. These are the main reasons for the appearance of the present book. As part of the textbook *Modern Electrochemistry 1 Ionics*, *A Guide to Problems in Modern Electrochemistry Part 1 Ionics* compiles many of the solutions to the exercises and problems presented in the text as well as many new problems.

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