

Chem 225 Analytical Chemistry 1

Problem Set No. 1_2

Redox Titrations

1. Selenium in a 10.0-g soil sample is distilled as the tetrabromide, which is collected in aqueous solution where it is hydrolyzed to SeO_3^{2-} . The SeO_3^{2-} is determined iodometrically, requiring 4.5 mL of standard thiosulfate solution for the titration. If the thiosulfate titer is 0.049 mg $\text{K}_2\text{Cr}_2\text{O}_7/\text{mL}$, what is the concentration of selenium in the soil in ppm?
2. The calcium in a 5.00-mL serum sample is precipitated as CaC_2O_4 with ammonium oxalate. The filtered precipitate is dissolved in acid, the solution is heated, and the oxalate is titrated with 0.00100 M KMnO_4 requiring 4.94 mL. Calculate the concentration of calcium in the serum in meq/L (equivalents based on charge).
3. A 2.50-g sample containing As_2O_3 , Na_2HASO_3 , and inert material is dissolved, and the pH is adjusted to neutral with excess NaHCO_3 . The As(III) is titrated with 0.150 M I_2 solution, requiring 11.3 mL to just reach the end point. Then, the solution (all the arsenic in the +5 state now) is acidified with HCl, excess KI is added, and the liberated I_2 is titrated with 0.120 M $\text{Na}_2\text{S}_2\text{O}_3$, requiring 41.2 mL. Calculate the percent As_2O_3 and Na_2HASO_3 in the sample.
4. If 1.00 mL KMnO_4 solution will react with 0.125 g Fe^{2+} and if 1.00 mL $\text{KHC}_2\text{O}_4 \cdot \text{H}_2\text{C}_2\text{O}_4$ solution will react with 0.175 mL of the KMnO_4 solution, how many milliliters of 0.200 M NaOH will react with 1.00 mL of the tetroxalate solution? (All three protons on the tetroxalate are titratable.)
5. The sulfide content in a pulp plant effluent is determined with a sulfide ion- selective electrode using the method of standard additions for calibration. A 10.0-mL sample is diluted to 25.0 mL with water and gives a potential reading of -216.4 mV. A similar 10.0-mL sample plus 1.00 mL of 0.030 M sulfide standard diluted to 25.0 mL gives a reading of -224.0 mV. Calculate the concentration of sulfide in the sample.

Redox Problem Set 1 Reactions And Stoichiometry

**Frank Schaffer Publications, Joan
DiStasio**



Redox Problem Set 1 Reactions And Stoichiometry:

Chemical Processes For Environmental Engineering Teh Fu Yen,2007-02-06 This book deals with basic principles such as chemical equilibrium and chemical processes concepts which make up the basic tools necessary to design a more efficient system to solve environmental problems Useful as a textbook for both graduate and undergraduate the material also serves as an excellent source for professional research in the field of environmental engineering or environmental science a

Bioreaction Engineering Principles John Villadsen,Jens Nielsen,Gunnar Lidén,2011-07-12 The present text is a complete revision of the 2nd edition from 2003 of the book with the same title In recognition of the fast pace at which biotechnology is moving we have rewritten several chapters to include new scientific progress in the field from 2000 to 2010 More important we have changed the focus of the book to support its use not only in universities but also as a guide to design new processes and equipment in the bio industry A new chapter has been included on the prospects of the bio refinery to replace many of the oil and gas based processes for production of especially bulk chemicals This chapter also serves to make students in Chemical Engineering and in the Bio Sciences enthusiastic about the whole research field As in previous editions we hope that the book can be used as textbook for classes even at the undergraduate level where chemical engineering students come to work side by side with students from biochemistry and microbiology To help the chemical engineering students Chapter 1 includes a brief review of the most important parts of microbial metabolism In our opinion this review is sufficient to understand microbial physiology at a sufficiently high level to profit from the rest of the book Likewise the bio students will not be overwhelmed by mathematics but since the objective of the book is to teach quantitative process analysis and process design at a hands on level some mathematics and model analysis is needed We hope that the about 100 detailed examples and text notes together with many instructive problems will be sufficient to illustrate how model analysis is used also in Bio reaction Engineering

Chemistry James E. Brady,Fred Senese,2004-02-04 Publisher Description *Chemical Problem-solving by Dimensional Analysis* Arnold B. Loebel,1978

Chemistry Homework Frank Schaffer Publications,Joan DiStasio,1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more

White's Handbook of Chlorination and Alternative Disinfectants Black & Veatch Corporation,2011-09-20 New edition covers the latest practices regulations and alternative disinfectants Since the publication of the Fourth Edition of White s Handbook of Chlorination and Alternative Disinfectants more than ten years ago the water industry has made substantial advances in their understanding and application of chlorine hypochlorite and alternative disinfectants for water and wastewater treatment This Fifth Edition with its extensive updates and revisions reflects the current state of the science as well as the latest practices Balancing theory with practice the Fifth Edition covers such important topics as Advances in the use of UV and ozone as disinfectants Alternative disinfectants such as chlorine dioxide iodine and bromine related products Advanced oxidation processes for drinking water and wastewater treatment New developments and information for

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Fundamental Bioengineering John Villadsen, 2016-03-01 A thorough introduction to the basics of bioengineering with a focus on applications in the emerging white biotechnology industry As such this latest volume in the Advanced Biotechnology series covers the principles for the design and analysis of industrial bioprocesses as well as the design of bioremediation systems and several biomedical applications No fewer than seven chapters introduce stoichiometry kinetics thermodynamics and the design of ideal and real bioreactors illustrated by more than 50 practical examples Further chapters deal with the tools that enable an understanding of the behavior of cell cultures and enzymatically catalyzed reactions while others discuss the analysis of cultures at the level of the cell as well as structural frameworks for the successful scale up of bioreactions In addition a short survey of downstream processing options and the control of bioreactions is given With contributions from leading experts in industry and academia this is a comprehensive source of information peer reviewed by experts in the field

Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

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solving A wide variety of case studies and quantitative problems accompany each chapter giving each instructor the flexibility to tailor the material to his her course Many problems have no single correct answer illustrating the analytical nature of solving real world environmental problems Materials and Applications for Sensors and Transducers III
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