

# 1 Atoombouw en periodiek systeem

## OPDRACHTEN

1

### Atoombouw

Maak bij deze opdracht gebruik van Binas tabel 25A en 99. Leg steeds uit hoe je aan je antwoord komt.

- Hoeveel protonen heeft een neonatoom?  
Het atoomnummer van neon is 10. Neon heeft dus tien protonen.
- Hoeveel elektronen heeft een zilveratoom?  
Het atoomnummer van zilver is 47. Een zilveratoom heeft dus 47 protonen en ook 47 elektronen.
- Hoeveel neutronen heeft O-18?  
Het atoomnummer van zuurstof is 8, het massagetal van O-18 is 18. Het aantal neutronen is gelijk aan het massagetal minus het atoomnummer:  
 $18 - 8 = 10$  neutronen.
- Van welk(e) element(en) bestaan atomen met 23 protonen?  
Het aantal protonen is gelijk aan het atoomnummer. Elk element heeft een uniek atoomnummer. Atoomnummer 23 is van vanadium, V.
- Van welk(e) element(en) bestaan atomen met massagetal 109?  
Omdat het aantal neutronen in een kern kan variëren, bestaan er van de meeste elementen meerdere isotopen met verschillende massagetallen. Verschillende elementen kunnen dus ook atomen hebben met hetzelfde massagetal. Opzoeken in Binas tabel 25A wijst uit dat er van zilver en cadmium (Ag en Cd) isotopen met massagetal 109 bestaan.

2

### Periodiek systeem

In het periodiek systeem zijn de elementen in groepen en perioden gerangschikt. Gebruik bij deze opdracht Binas tabel 99.

- Geef de symbolen van alle elementen in groep 10.  
Ni, Pd, Pt, Ds
- Geef de symbolen van alle elementen in periode 3.  
Na, Mg, Al, Si, P, S, Cl, Ar
- Hoe heten de elementen in groep 17?  
halogenen
- In welke groep staan de edelgassen?  
groep 18
- Welke periode bevat geen metalen?  
periode 1

3

### Isotopen

Om aan te geven welke specifieke isotoop wordt bedoeld, zijn drie notaties in symbolen mogelijk.

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## **Table of Contents Nova Scheikunde 4 Vwo**

1. Understanding the eBook Nova Scheikunde 4 Vwo
  - The Rise of Digital Reading Nova Scheikunde 4 Vwo
  - Advantages of eBooks Over Traditional Books
2. Identifying Nova Scheikunde 4 Vwo
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Nova Scheikunde 4 Vwo
  - User-Friendly Interface
4. Exploring eBook Recommendations from Nova Scheikunde 4 Vwo
  - Personalized Recommendations
  - Nova Scheikunde 4 Vwo User Reviews and Ratings
  - Nova Scheikunde 4 Vwo and Bestseller Lists
5. Accessing Nova Scheikunde 4 Vwo Free and Paid eBooks

- Nova Scheikunde 4 Vwo Public Domain eBooks
  - Nova Scheikunde 4 Vwo eBook Subscription Services
  - Nova Scheikunde 4 Vwo Budget-Friendly Options
6. Navigating Nova Scheikunde 4 Vwo eBook Formats
- ePub, PDF, MOBI, and More
  - Nova Scheikunde 4 Vwo Compatibility with Devices
  - Nova Scheikunde 4 Vwo Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Nova Scheikunde 4 Vwo
  - Highlighting and Note-Taking Nova Scheikunde 4 Vwo
  - Interactive Elements Nova Scheikunde 4 Vwo
8. Staying Engaged with Nova Scheikunde 4 Vwo
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Nova Scheikunde 4 Vwo
9. Balancing eBooks and Physical Books Nova Scheikunde 4 Vwo
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Nova Scheikunde 4 Vwo
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Nova Scheikunde 4 Vwo
- Setting Reading Goals Nova Scheikunde 4 Vwo
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Nova Scheikunde 4 Vwo
- Fact-Checking eBook Content of Nova Scheikunde 4 Vwo
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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