

Mass of neutron = mass of proton

Mass of electron = $\frac{1}{1840}$ mass of proton

Diameter of nucleus is of the order of 10^{-14} m

Diameter of orbits = 10^4 times dia of molecule

Diameter of electron = 10^{-15} m

Charge on electron = -1.602×10^{-19} coulomb

Charge on proton = $+1.602 \times 10^{-19}$ coulomb

Objective Electrical

1.2. UNIT OF CURRENT

The charge on an electron is measured in terms of coulomb. The unit of current is coulomb per second and is called ampere. Thus

$$I \text{ (Ampere)} = \frac{\text{coulomb}}{\text{second}} = \frac{\Delta q}{\Delta t}$$

One coulomb is equivalent to the charge of 6.28×10^{18} electrons.

One ampere = 10^{10} electrons per second

1.3. ELECTROMOTIVE FORCE

Electromotive force or potential of a body is the work done in joules to bring a unit electric charge from infinity to the body. It is expressed in terms of volts.

The *potential difference* is defined as that which causes current to flow in the closed circuit.

1.4. RESISTANCE

Resistance is the property of a substance due to which it opposes the flow of electrons (i.e., electric current) through it. The unit of resistance is ohm (Ω).

Metals, acids and salt solutions are good conductors of electricity. Silver, copper and aluminium offer least resistance to flow of current and are called very good conductor of electricity. The electrons while flowing through the molecules or the atoms of the conductor, collide with other atoms and electrons, thereby producing heat.

Objective Of Electrical Engineering By Handa

Shasha Hu



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