

## chemistry stoichiometry questions and answers

### University of Binawan

1. Complete combustion of ammonia gas requires oxygen gas with equality reaction :



With Assuming that the oxygen level in the air is 20%, then the volume of air needed to burn 4 liters of ammonia gas is...

- A. 5 liters
- B. 7 liters
- C. 10 liters
- D. 35 liters

**Answer: D**

#### Discussion

Equality reaction equivalent :



Volume gas oxygen =  $(7/4) \times 4$  liters = 7 liters

Volume oxygen = 20% volume air

7 liters = 0.2 volume of air

volume udara =  $7/0.2 = 35$  litres

2. Every 1 liter of nitrogen gas is appropriate finished react with 2.5 liters of oxygen gas to form 1 liter of nitrogen oxide gas . If the volume is measured at the same temperature and pressure , then formula molecule the nitrogen oxide is ....

- A.  $\text{N}_2\text{O}$
- B.  $\text{NO}$
- C.  $\text{N}_2\text{O}_3$  \_ \_ \_
- D.  $\text{N}_2\text{O}_5$  \_ \_

**Answer: D**

#### Discussion

Equality reaction :



The number of N atoms before reaction = Number of N atoms after reaction

$$4 = 2x$$

$$x = 2$$

Number of O atoms before reaction = Number of O atoms after reaction

$$10 = 2y$$

$$y = 5$$

So, formula molecule The nitrogen oxide is  $\text{N}_2\text{O}_5$ .

3. Propane gas ( $\text{C}_3\text{H}_8$ ) is burned perfect with reaction :



For burn perfect every liters of propane gas, oxygen gas is required as much .....

- A. 1.0 liters
- B. 3.0 liters

# Stoichiometry Questions Answers

**Jicheng Xie**



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## **Table of Contents Stoichiometry Questions Answers**

1. Understanding the eBook Stoichiometry Questions Answers
  - The Rise of Digital Reading Stoichiometry Questions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Stoichiometry Questions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Stoichiometry Questions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Stoichiometry Questions Answers
  - Personalized Recommendations
  - Stoichiometry Questions Answers User Reviews and Ratings
  - Stoichiometry Questions Answers and Bestseller Lists
5. Accessing Stoichiometry Questions Answers Free and Paid eBooks
  - Stoichiometry Questions Answers Public Domain eBooks
  - Stoichiometry Questions Answers eBook Subscription Services
  - Stoichiometry Questions Answers Budget-Friendly Options

6. Navigating Stoichiometry Questions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Stoichiometry Questions Answers Compatibility with Devices
  - Stoichiometry Questions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Stoichiometry Questions Answers
  - Highlighting and Note-Taking Stoichiometry Questions Answers
  - Interactive Elements Stoichiometry Questions Answers
8. Staying Engaged with Stoichiometry Questions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Stoichiometry Questions Answers
9. Balancing eBooks and Physical Books Stoichiometry Questions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Stoichiometry Questions Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Stoichiometry Questions Answers
  - Setting Reading Goals Stoichiometry Questions Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Stoichiometry Questions Answers
  - Fact-Checking eBook Content of Stoichiometry Questions Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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