

Name : \_\_\_\_\_

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# Acids and Bases Strength

Write True (T) or False (F).

- i. Stronger acids have weaker conjugate bases and weaker acids have stronger conjugate bases. \_\_\_\_\_
- ii. Stronger bases have stronger conjugate acids and weaker bases have weaker conjugate acids. \_\_\_\_\_

Write equations that show how the following reactions occur in water.

- i.  $F^-$  acts as a base. \_\_\_\_\_
- ii.  $HNO_2$  acts as an acid. \_\_\_\_\_
- iii.  $Fe(H_2O)_6^{3+}$  acts as an acid. \_\_\_\_\_
- iv.  $HCO_3^-$  acts as a base. \_\_\_\_\_
- v.  $HCO_3^-$  acts as an acid. \_\_\_\_\_
- vi.  $Al(H_2O)_5(OH)^{2+}$  acts as a base. \_\_\_\_\_

What volume of 0.45 M LiOH would be needed to neutralize 60.0 mL of 0.15 M HI?

Calculate the pH values of the following solutions of strong acids and bases.

i. 0.001 M HCl

ii. 0.76 M KOH

iii.  $2.8 \times 10^{-4}$  M  $Ba(OH)_2$

# Strength Of Acids And Bases Worksheet Answers

**Tobias Bleicker**



**Strength Of Acids And Bases Worksheet Answers:**

The Enigmatic Realm of **Strength Of Acids And Bases Worksheet Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Strength Of Acids And Bases Worksheet Answers** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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