

## Student Exploration: pH Analysis

**Essential** **Learning Objectives** **Students Will Be Able to**

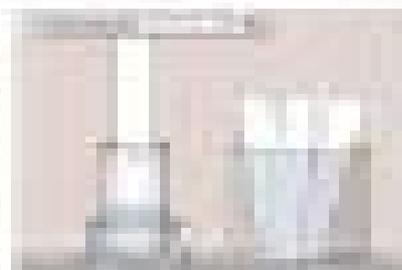
**Write Knowledge Statements** (Students will be able to write the following.)

- Water** can self-ionize and produce hydrogen ions ( $H^+$ ) and hydroxide ions ( $OH^-$ ) in equal amounts at all times.
  - Water is neutral (pH=7.0) when
    - there is an equal number of hydrogen ions and hydroxide ions.
  - Water is acidic when there is a greater number of hydrogen ions than hydroxide ions.
  - Water is basic when there is a greater number of hydroxide ions than hydrogen ions.
- Water** is a weak acid that produces hydrogen ions ( $H^+$ ) when dissolved in water. Water does not ionize at all times.
  - Water is neutral (pH=7.0) when
    - there is an equal number of hydrogen ions and hydroxide ions.
  - Water is acidic when there is a greater number of hydrogen ions than hydroxide ions.
  - Water is basic when there is a greater number of hydroxide ions than hydrogen ions.

### Essential Vocabulary

The amount of an acid or base is measured on the **pH scale**. The amount of **acid** is called the **concentration** of hydrogen ions. It is a measure of how many hydrogen ions are in a solution. The pH scale goes from zero (0) with a high concentration of hydrogen ions to fourteen (14) with a low concentration of hydrogen ions. **Neutral** substances have a pH of 7, which means there is an equal number of hydrogen ions and hydroxide ions.

**Acidic** substances have a greater number of hydrogen ions than hydroxide ions. **Basic** substances have a greater number of hydroxide ions than hydrogen ions.



- Neutral** substances have a pH of 7.0.
  - Neutral substances have a greater number of hydrogen ions than hydroxide ions.
  - Neutral substances have a greater number of hydroxide ions than hydrogen ions.
- Acidic** substances have a pH less than 7.0.
  - Acidic substances have a greater number of hydrogen ions than hydroxide ions.
  - Acidic substances have a greater number of hydroxide ions than hydrogen ions.
- Basic** substances have a pH greater than 7.0.
  - Basic substances have a greater number of hydrogen ions than hydroxide ions.
  - Basic substances have a greater number of hydroxide ions than hydrogen ions.

# Student Exploration Ph Analysis Answer Key

**J Ma**



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