

1 • Matter and Measurement

2 100 BP 1F 212 of degrees are smaller than 100K
0 FP 32
PRACTICE TEST

- Which physical state of matter exhibits the greatest change in volume with changes in temperature or pressure? *mostly empty space*
 a) solid b) liquid **c) gas**
easy to compress
- According to the Kinetic Molecular Theory particles of a solid *True*
 a) are bound in a regular array and do not move.
 b) float freely within an array occupying various positions relative to neighbors. *liquid*
 c) have no relationship to the microscopic structure of the solid.
d) vibrate back and forth but do not move past immediate neighbors.
 e) float freely in the inside but do not move on the surface.
- What volume of a liquid having a density of 1.48 g/cm³ is needed to supply 5.00 grams of the liquid?
 a) 0.296 cm³ **d) 3.38 cm³**
 b) 1.48 cm³ e) 7.40 cm³
 c) 2.26 cm³ *5.00g * 1 cm³ / 1.48g =*
- The density of aluminum is 2.70 g/cm³. If a cube of aluminum weighs 13.5 grams, what is the length of the edge of the cube?
 a) 5.00 cm d) 0.312 cm
b) 1.71 cm e) 0.200 cm
 c) 1.25 cm
*13.5g * 1 cm³ / 2.70g = 5.00 cm³*
∛5.00 cm =

- Which temperature change is the smallest?
 a) 10°C to 20°C **d) 10°F to 20°C**
~~b) 10 K to 20°C~~ **e) 10°F to 20°F**
 c) 10 K to 20 K
- The number of significant figures in 0.06060 x 10⁻⁵ is
 a) 2 d) 5
 b) 3 e) 6
c) 4
- The number, three hundred and fifty thousand, written in scientific notation is best written as
 a) 350 d) 3.50 x 10⁵
 b) 3.5 x 10⁶ e) 3.50 x 10⁻⁵
c) 3.5 x 10⁵ *assume least # of digits significant digits*
350,000 = 3.5 x 10⁵
- The mass of a sample weighted on an electronic balance that is sensitive to ±0.3 mg is 1.2300 g. The number of significant figures in this measurement is *.3mg = .0003g*
 a) 1 d) 4
 b) 2 **c) 5** *1.2300 ± .0003*
 e) 3
- What is the numerical value of:
 1.5 cm - 7.222 x 10⁻¹ cm?
 a) 0.7778 cm **d) 0.8 cm**
 b) 0.778 cm e) 7.072 x 10⁻¹ cm
 c) 0.78 cm

$$\begin{array}{r} 1.5 \text{ cm} \\ - 7.222 \text{ cm} \\ \hline .7778 \text{ cm} \\ \hline .8 \text{ cm} \end{array}$$

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Zhe Xu

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Table of Contents Ap Chemistry Chapter 1 Test

1. Understanding the eBook Ap Chemistry Chapter 1 Test
 - The Rise of Digital Reading Ap Chemistry Chapter 1 Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Ap Chemistry Chapter 1 Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ap Chemistry Chapter 1 Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ap Chemistry Chapter 1 Test
 - Personalized Recommendations
 - Ap Chemistry Chapter 1 Test User Reviews and Ratings
 - Ap Chemistry Chapter 1 Test and Bestseller Lists
5. Accessing Ap Chemistry Chapter 1 Test Free and Paid eBooks
 - Ap Chemistry Chapter 1 Test Public Domain eBooks
 - Ap Chemistry Chapter 1 Test eBook Subscription Services
 - Ap Chemistry Chapter 1 Test Budget-Friendly Options

6. Navigating Ap Chemistry Chapter 1 Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Ap Chemistry Chapter 1 Test Compatibility with Devices
 - Ap Chemistry Chapter 1 Test Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ap Chemistry Chapter 1 Test
 - Highlighting and Note-Taking Ap Chemistry Chapter 1 Test
 - Interactive Elements Ap Chemistry Chapter 1 Test
8. Staying Engaged with Ap Chemistry Chapter 1 Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ap Chemistry Chapter 1 Test
9. Balancing eBooks and Physical Books Ap Chemistry Chapter 1 Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ap Chemistry Chapter 1 Test
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ap Chemistry Chapter 1 Test
 - Setting Reading Goals Ap Chemistry Chapter 1 Test
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ap Chemistry Chapter 1 Test
 - Fact-Checking eBook Content of Ap Chemistry Chapter 1 Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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