

CAPE CHEMISTRY



UNIT 2-PAPER 2-MODULE 1

STRUCTURAL ISOMERISM

JUNE 2024 EXAM SOLUTIONS

Unit 2 Paper 2

Raleigh Schorling, John Roscoe Clark

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 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{S}$
 $4\text{CH}_4 + \text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$
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 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 5\text{NF}_3 + 3\text{SNF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{S} + 6\text{NaBr} + 3\text{SNO}_2 + 3\text{SNF}_2 + \text{N}_2\text{O}_3$ using the

following equation $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 3 $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 4 $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3 \text{NaBr}$
 5 $3 \text{SnO}_2 + \text{NF}_3 \rightarrow 3 \text{SnF}_2 + \text{N}_2\text{O}_3$
 using the following equation
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